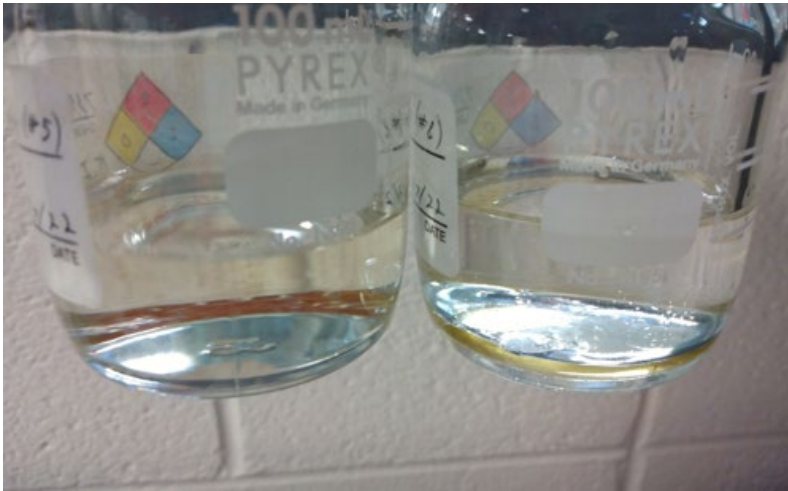


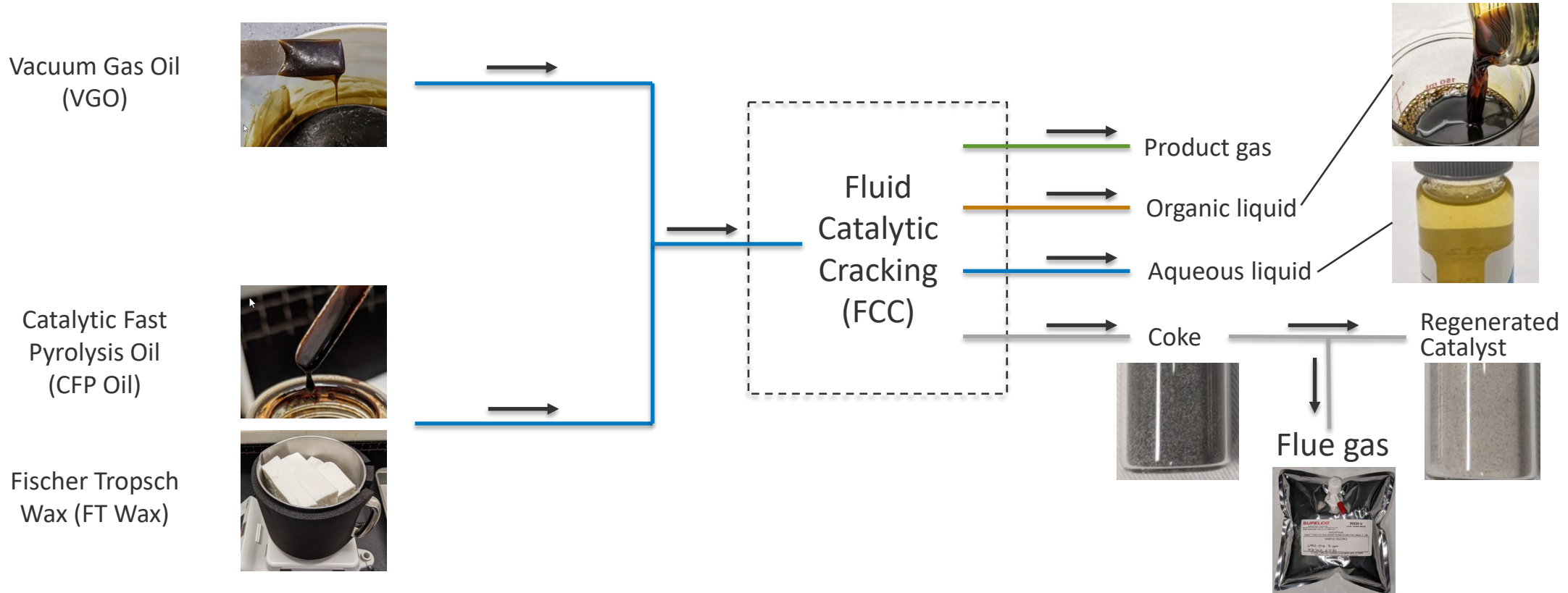
Biocarbon Tracking in FCC Co-processing of Biogenic Feedstocks



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Braden Peterson, Rebecca
Jackson, Earl Christensen, Kim
Magrini, and Robert Baldwin

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Co-processing of Biogenic Fuels with VGO



- Measure biocarbon in liquid product to determine biocarbon incorporation.
- Track biocarbon across other output fractions.

Similar study: G. Fogassy et al., The fate of bio-carbon in FCC co-processing products, Green Chemistry 14(5) (2012).

Definitions

¹⁴C – isotope

Half life of 5730 years.

Appears as a small fraction (1e-12) in biogenic feedstocks.

AMS – Accelerated Mass Spectrometer

Measures ¹⁴C/¹²C ratio.

pMC – percent Modern Carbon

Referenced to biogenic material in 1950.

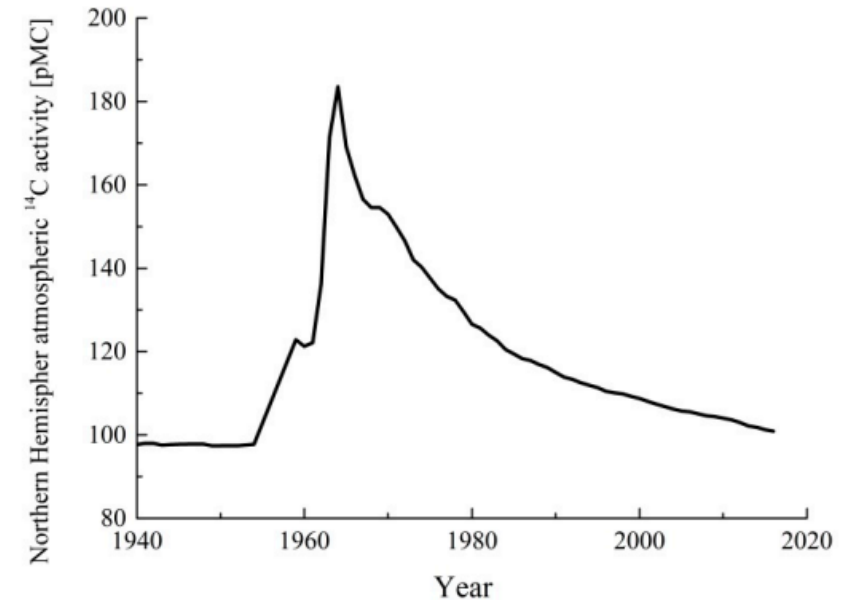
Can be above 100% depending on time of growth of biomass.

Percent biogenic carbon (percent biocarbon)

Normalized between fossil feed sample (0% biocarbon) and biogenic feed sample (100% biocarbon).

Example:

	pMC	% biocarbon
CFP Oil	102.90	100.00%
Liquid Product	5.66	5.33%
VGO	0.18	0.00%



S. Hammer, I. Levin, Monthly mean atmospheric D14CO₂ at Jungfrauoch and Schauinsland from 1986 to 2016, 2017

AMS Laboratories

Beta Analytic

Commercial Testing Laboratory in Florida

University of California Irvine

Keck – CCAMS Group

Earth System Science Department

pMC	UC Irvine	Beta Analytic	Theoretical Value
FT wax	83.30	83.70	
FT wax diluted with VGO	16.50		16.43
CFP Oil	102.90	102.40	
Co-processed FT wax	11.20	11.82	
Co-processed CFP oil	6.00	5.74	

$$pMC_{feed} = \frac{Y_{CFPO} \cdot Ccontent_{CFPO} \cdot pMC_{CFPO} + Y_{VGO} \cdot Ccontent_{VGO} \cdot pMC_{VGO}}{Y_{CFPO} \cdot Ccontent_{CFPO} + Y_{VGO} \cdot Ccontent_{VGO}}$$

pMC ... percent Modern Carbon (from ^{14}C analysis)

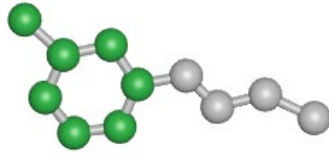
...alternatively also works for percent biocarbon

Y ...component mass fraction

$Ccontent$...from ultimate analysis

Carbon Scrambling

Fragments from biogenic molecules can attach to fossil molecules and vice versa. E.g., a molecule originating from biomass pyrolysis may pick up a short fossil carbon chain and produce a molecule in the jet fuel range that is now partially biogenic and partially fossil.



¹³C Study: C. Mukarakate et al., Isotopic Studies for Tracking Biogenic Carbon during Co-processing of Biomass and Vacuum Gas Oil, ACS Sustainable Chemistry & Engineering (2020).

Consider the extremes:

- No carbon scrambling, i.e., no biogenic carbon atoms combine with fossil molecules. Typical in decomposition reactions.
 - E.g., biogenic content in jet fuel shows that biogenic feed produced jet-fuel range molecules.
- Perfect carbon scrambling, i.e., all biogenic and fossil carbon atoms randomly exchange/combine. Typical in synthesis reactions.
 - All product fractions show the same biocarbon percentage as in the feed.

- Analysis of biogenic carbon incorporation into fuels is valid for any carbon scrambling.
- Following reactions paths via biocarbon analysis becomes less useful if carbon scrambling occurs.

Co-Processing of CFP Oil in Davison Circulating Riser

Key System Specifications

Feed Type: Liquids

Feed Rate: 1.2 L/hr (1.14 L/hr VGO, 60 mL/hr CFP Oil)

Catalyst Type: E-Cat.

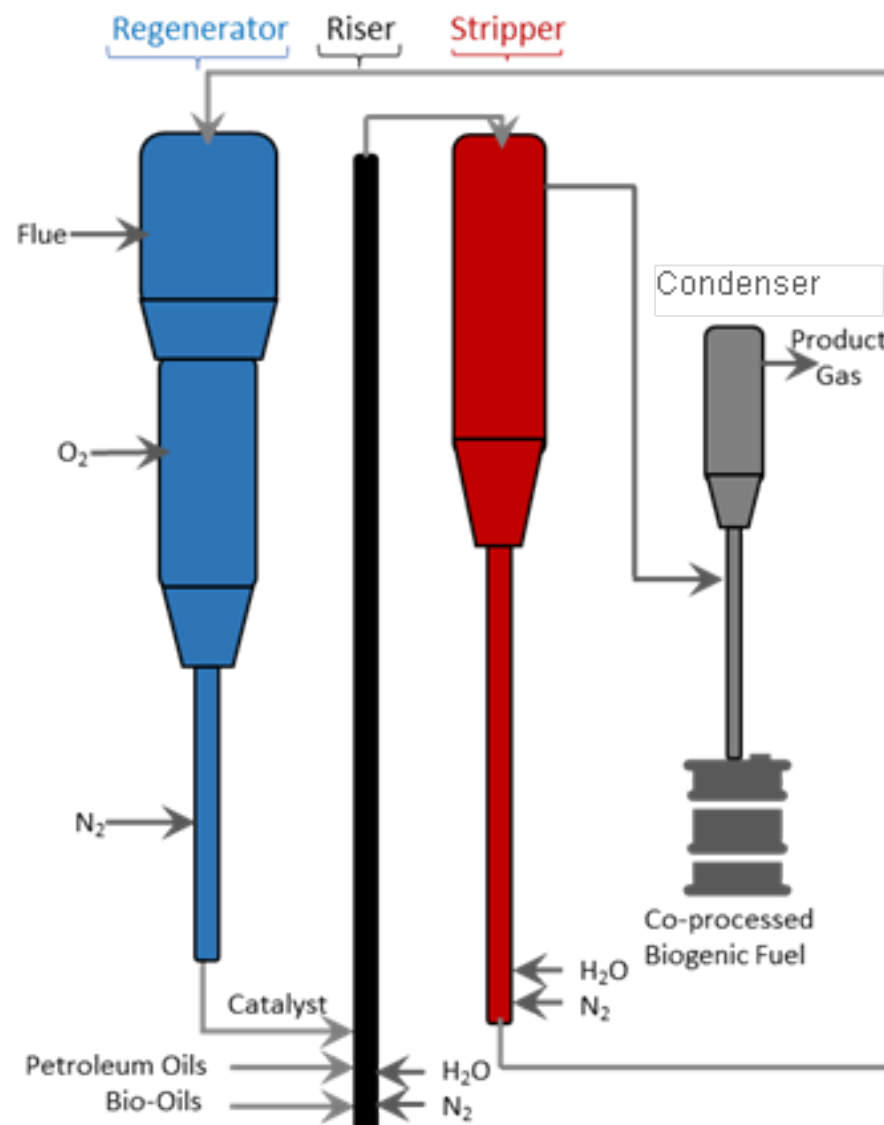
Catalyst Charge/Feed Rate: 1.8 kg, circulation rate 7-10 kg/hr

Operating Temperature Range: 521°C

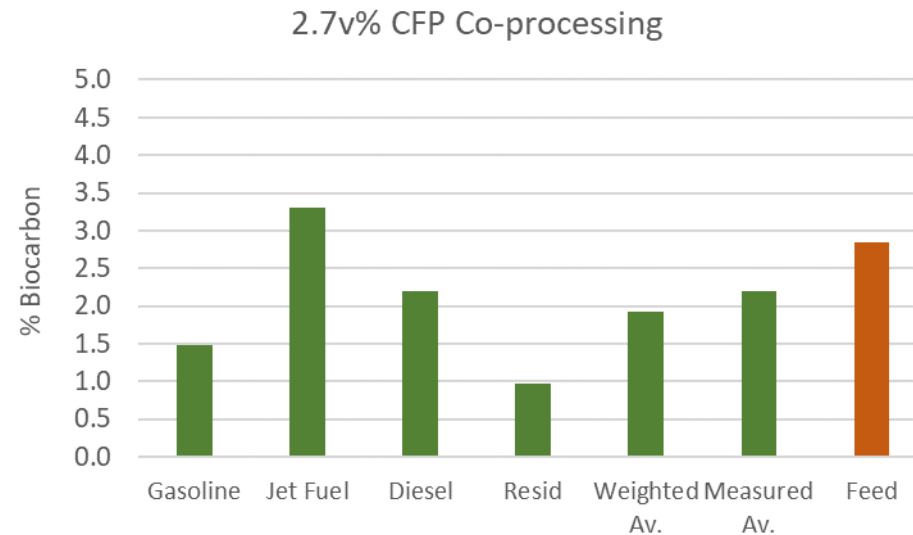
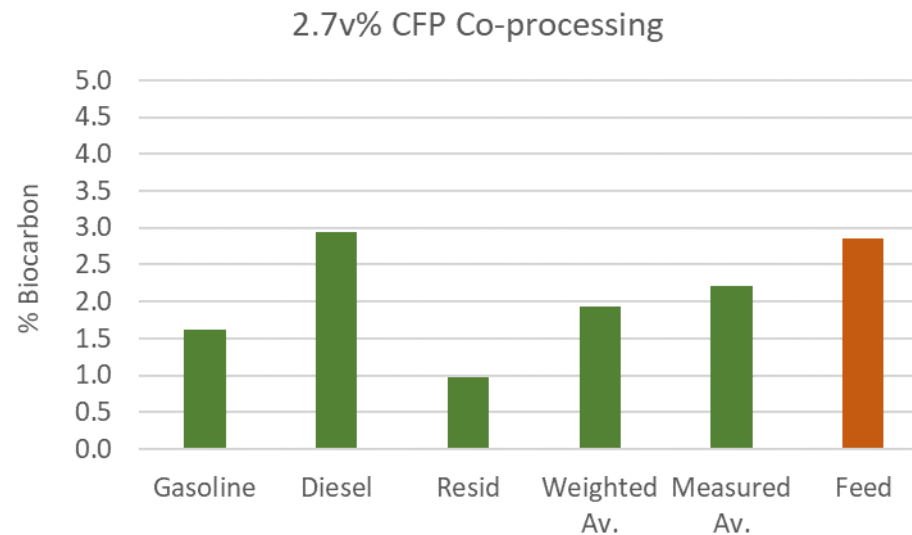
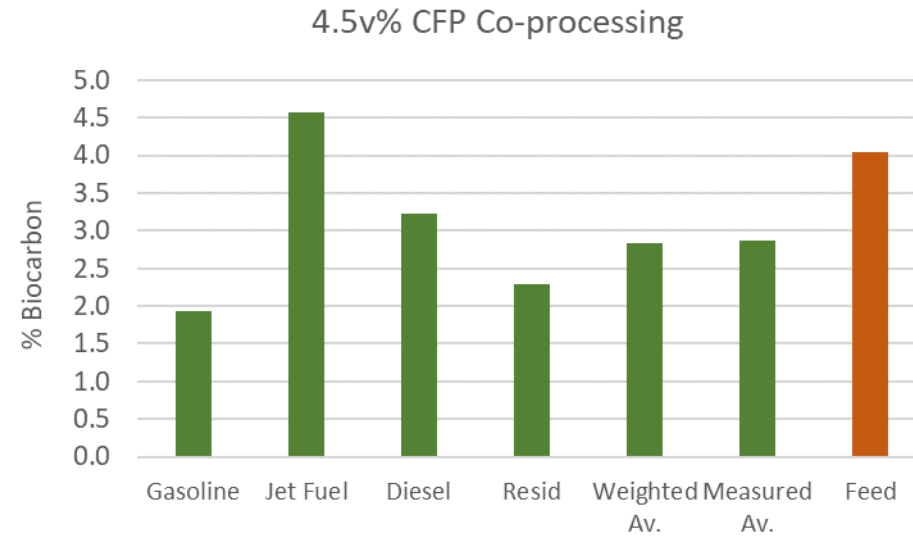
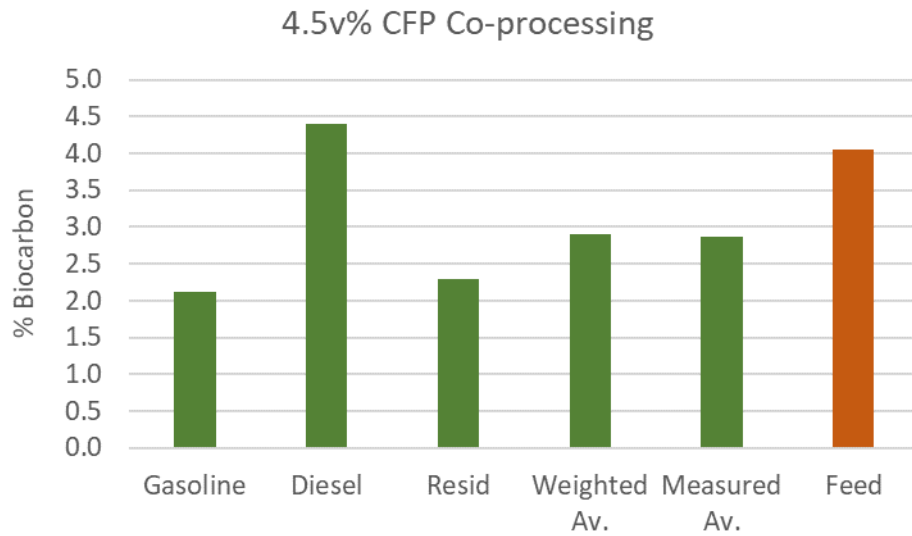
Operating Pressure Range: 2.6 bar

Biocarbon Analysis

- Liquid product and distilled fractions

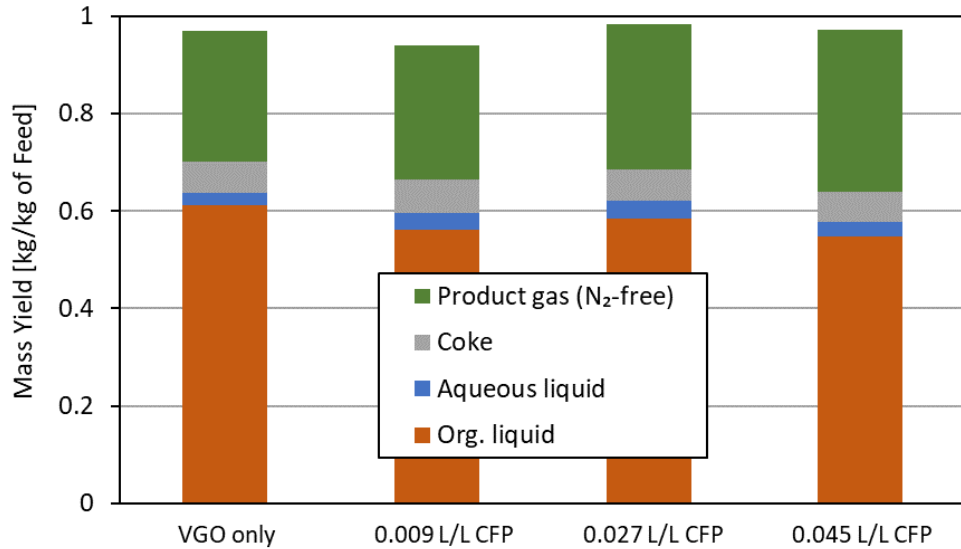


Biocarbon Measurements in Co-Processing Products

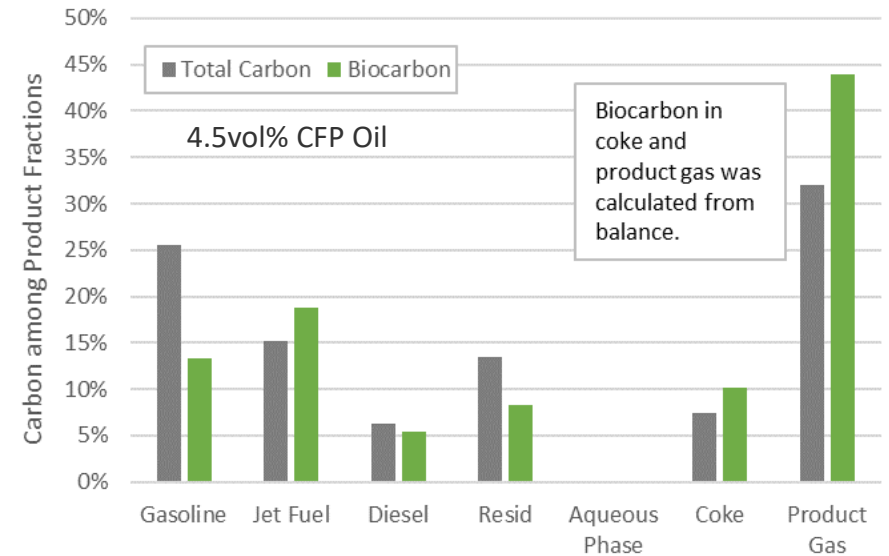


Biocarbon Incorporation into Liquid Fuels

Mass balance



Biocarbon Distribution
(as % of Biocarbon in Feed)



- A simple biocarbon incorporation into the liquid products can be calculated with the following equation:

$$\text{bioc incorporation into liquid} = \frac{\frac{\text{bioc}_{\text{product}}}{\text{bioc}_{\text{feed}}}}{\frac{\text{totalC}_{\text{product}}}{\text{totalC}_{\text{feed}}}} = \frac{\frac{\text{bioc}_{\text{product}}}{\text{totalC}_{\text{product}}}}{\frac{\text{bioc}_{\text{feed}}}{\text{totalC}_{\text{feed}}}} = e.g. \frac{44\%}{62\%} = \frac{2.8\%}{4.0\%} = 70\%$$

Co-processing of Fischer-Tropsch Wax in DCR

Key Specifications

Feed Type: 20w% FT wax with VGO

Catalyst Type: E-Cat.

Operating Temperature Range: 521°C

Operating Pressure Range: 2.6 bar

Biocarbon Analysis:

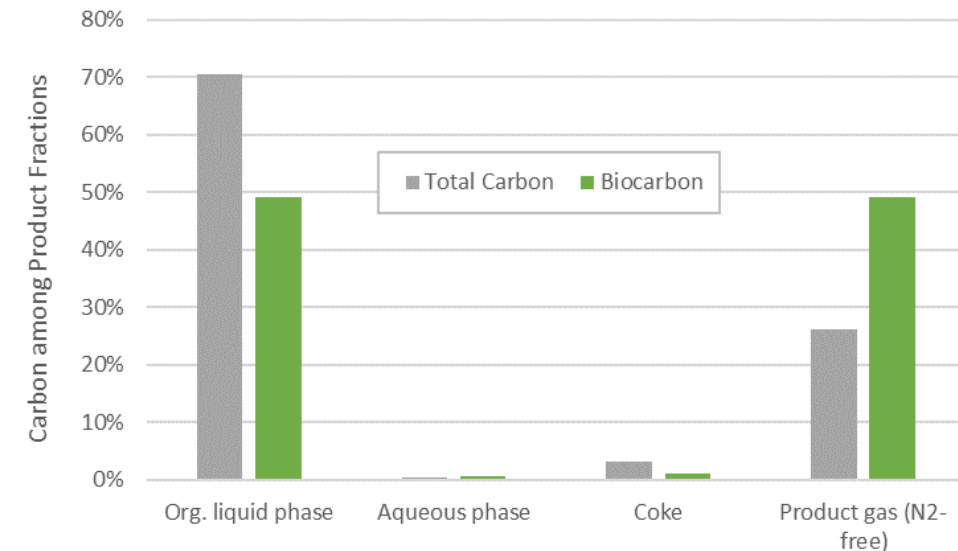
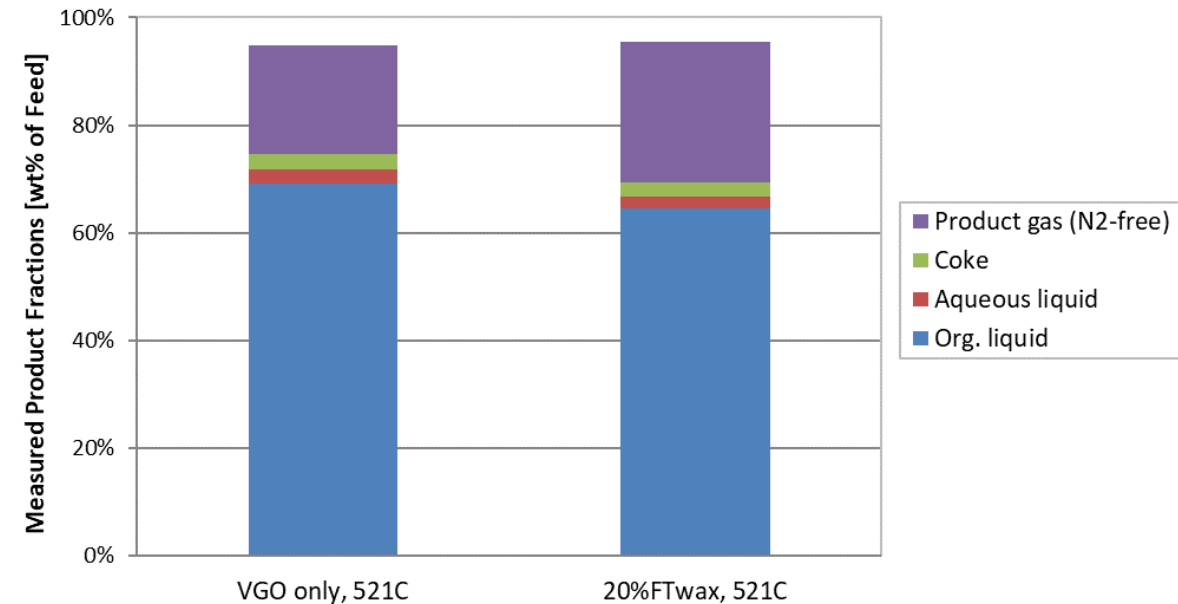
- Liquid product
- Gas bag from oxidized coke
- Catalyst after stripper

Results

Lower liquid production with co-processing of FT wax

Lower coke formation

- Gas bags showed 5.2% biogenic carbon (versus 16.3% in feed)
- Catalyst sample was too small to detect bio carbon



Co-process FT Wax in ACE Reactor

Advanced Cracking Evaluation Reactor – Kayser Technology

Co-processing of 40 wt% FT wax with 60% VGO

Temperature: 530°C

Submitted for Biocarbon Analysis:

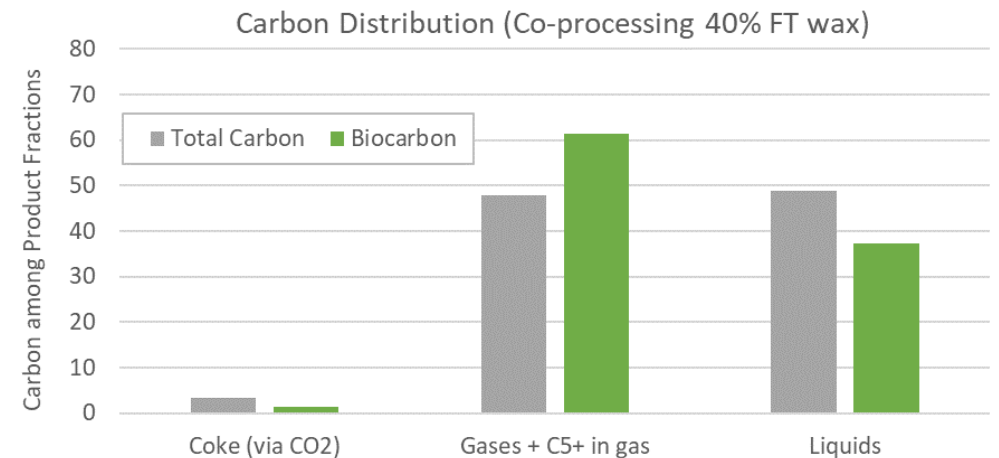
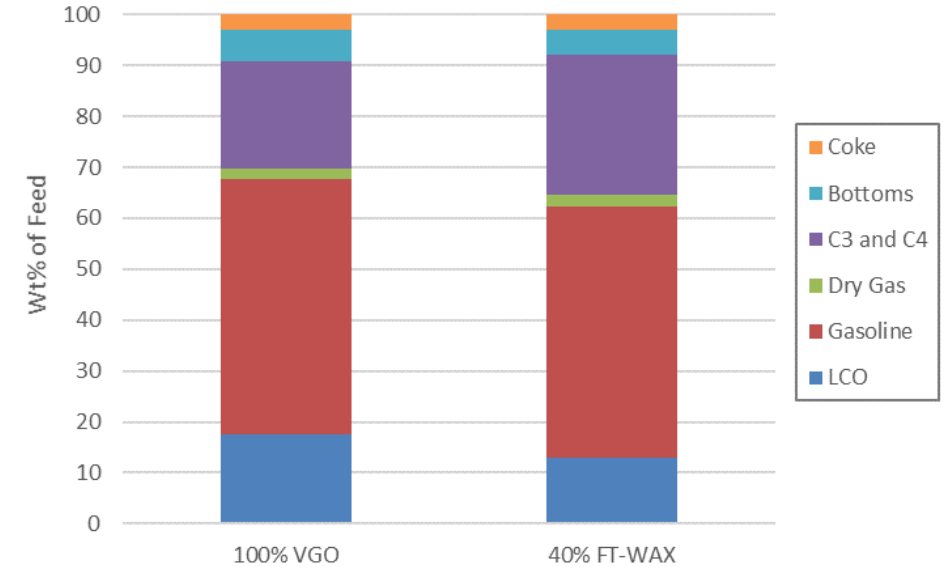
- Liquids
- Gas bags from coke oxidation
- Gas bags from product gas oxidation
 - Mix air with product gas (Lambda 4)
 - Flow mixture over CuO (600°C)
 - Methane remaining at below 0.01 mole%

Results

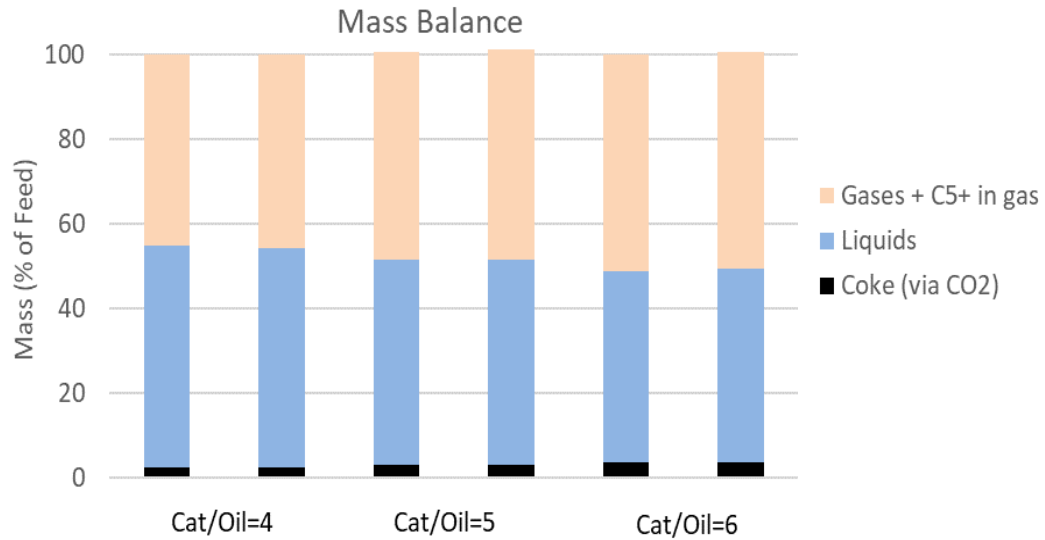
Lower diesel yields

Higher LPG yields (especially C₃ and C₄ olefins)

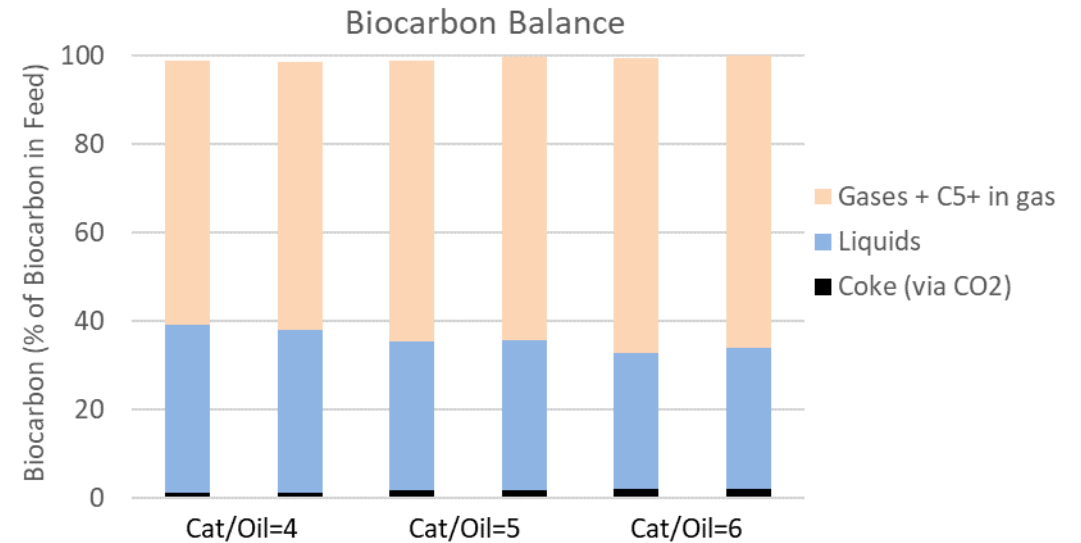
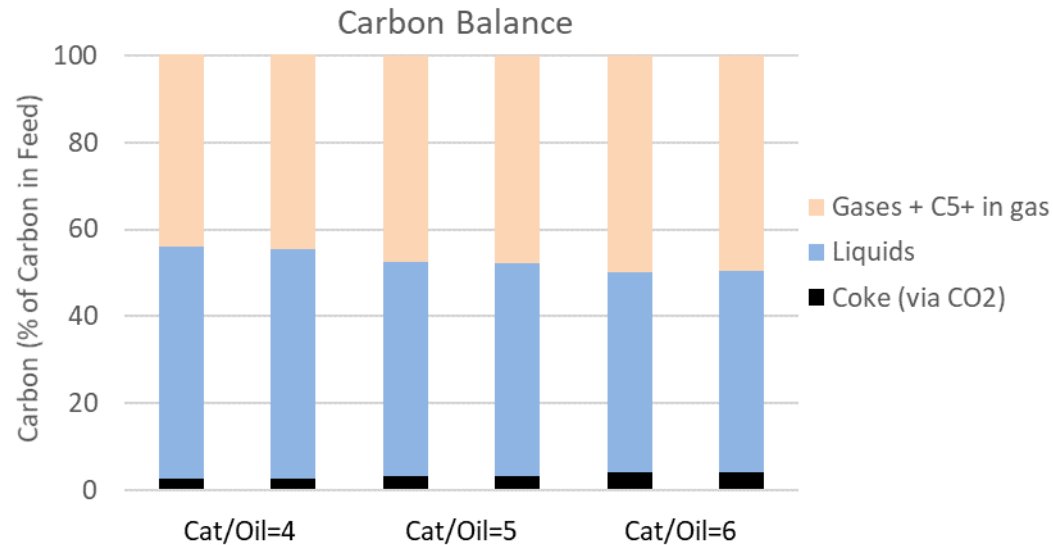
All product fractions confirmed by biocarbon analysis



ACE Reactor Results



- Experiments performed for three cat/oil ratios
- Experiments performed in duplicates
- Good mass, carbon, and biocarbon balances
- Biocarbon preferentially flows to gases



Summary

- AMS ^{14}C measurement generally provides good information for biocarbon balance.
- Some difficulties are observed for small biocarbon (<3%) or small carbon percentages (coke).
- ^{14}C measurement can confirm differences in liquid yields.
- On E-Cat, both CFP Oil and FT waxes cracked more easily than VGO and led to reduced liquid yields.
- Some fractions from distillation of products from co-processing CFP Oil received more biocarbon than others.



Thank You

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