



Heterodoxy in fast pyrolysis: Air-blown reactors and sugar production

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Orthodoxy in fast pyrolysis

- **Definition of pyrolysis:** “Thermal decomposition of organic compounds in the absence of oxygen.”
 - *Biorenewable Resources: Engineering New Products from Agriculture*
- **Products of pyrolysis:** “Bio-oil, biochar, non-condensable gases.”
 - *Biorenewable Resources: Engineering New Products from Agriculture*
- **Bio-oil:** “Bio-oil is an emulsion of lignin-derived oligomers in an aqueous phase composed primarily of carbohydrate-derived compounds.”
 - *Biorenewable Resources: Engineering New Products from Agriculture*

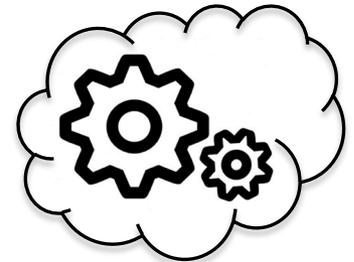
Heterodoxy in fast pyrolysis

- **Definition of pyrolysis:** “Thermal decomposition of organic compounds – and a little oxygen is just fine.”
- **Products of pyrolysis:** “Let’s add sugar among the major products.”
- **Bio-oil:** “Why would you even want to produce an unstable emulsion?”



A scientist and engineer walk into a bar...

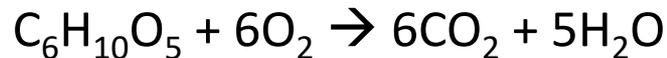
- **Scientist: Asks questions.**
- **Engineer: Solves problems.**



Why do we exclude oxygen from pyrolysis?

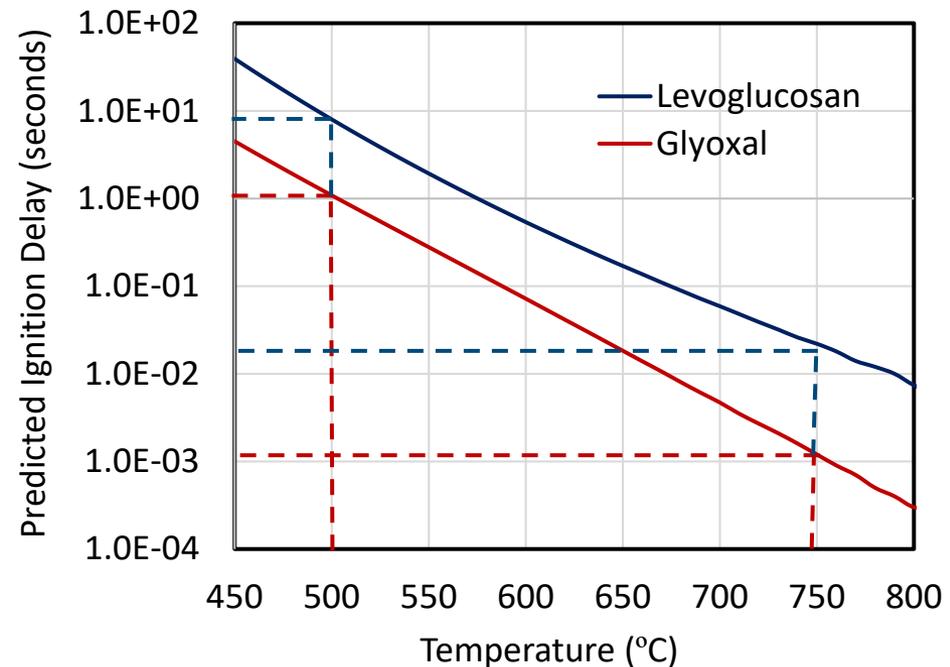


- Combustion happens:



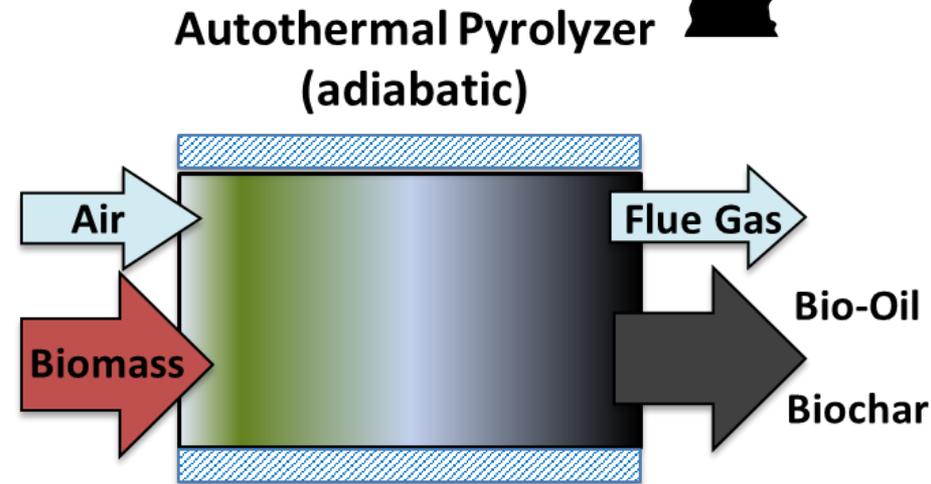
- Mitigating factors:

- Operation at low equivalence ratios
- Operation at temperatures lower than traditional combustion
- Variations in reaction rates among pyrolysis products



What is the upside of adding oxygen (as air)?

- Eliminates tail gas recycle
- Exothermic energy release of oxidation can provide enthalpy for pyrolysis
- Heat transfer bottleneck to process intensification can be removed!



$$\phi = \frac{Air}{Air_{Stoich}}$$

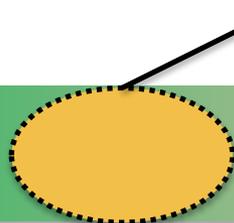
Autothermal Pyrolysis

$$0.06 \leq \phi \leq 0.12$$

Gasification

$$0.20 \leq \phi \leq 0.35$$

Pyrolysis
 $\phi = 0.00$

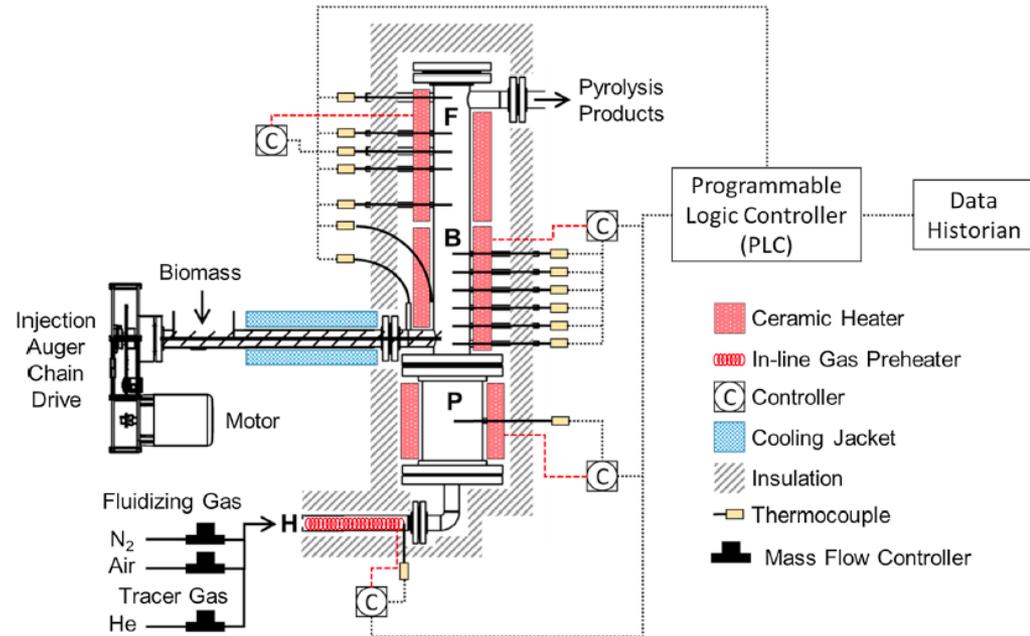


Combustion
 $\phi > 1.0$

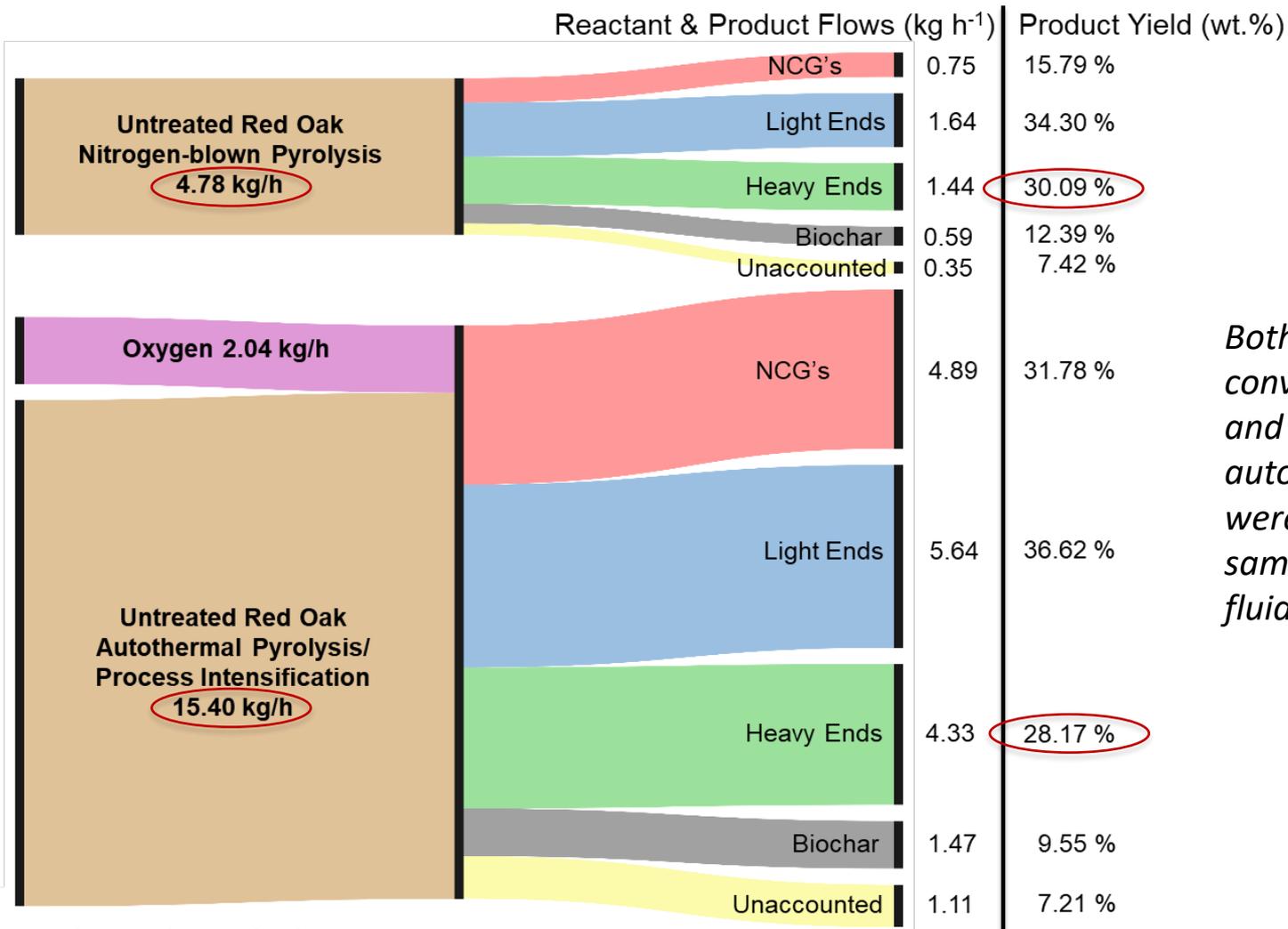
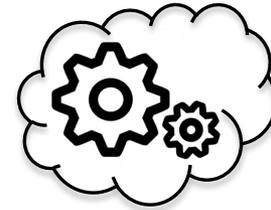
Design considerations for a lab-scale autothermal pyrolyzer



- Simulate adiabatic operation
 - Use of guard heaters to overcome parasitic heat losses at small scale
- Mitigate hot spots that promote undesired oxidation reactions
 - Use of fluidized bed for good internal heat and mass transfer
- Operate at minimum equivalence ratio that provides enthalpy for pyrolysis



Process intensification through autothermal pyrolysis



Both nitrogen-blown, conventional pyrolysis and air-blown ($\varphi=0.11$), autothermal pyrolysis were performed in the same 8.9 cm diameter fluidized bed reactor.

What is the impact of heat transfer on scaling up processes?



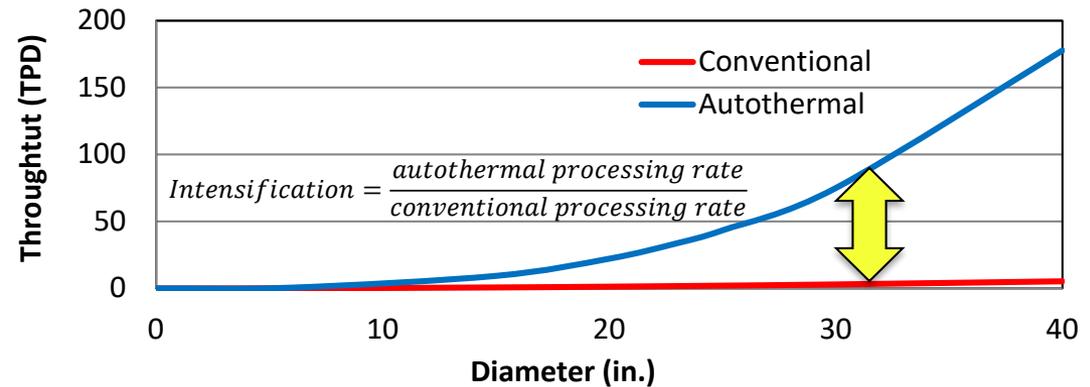
“hairless”



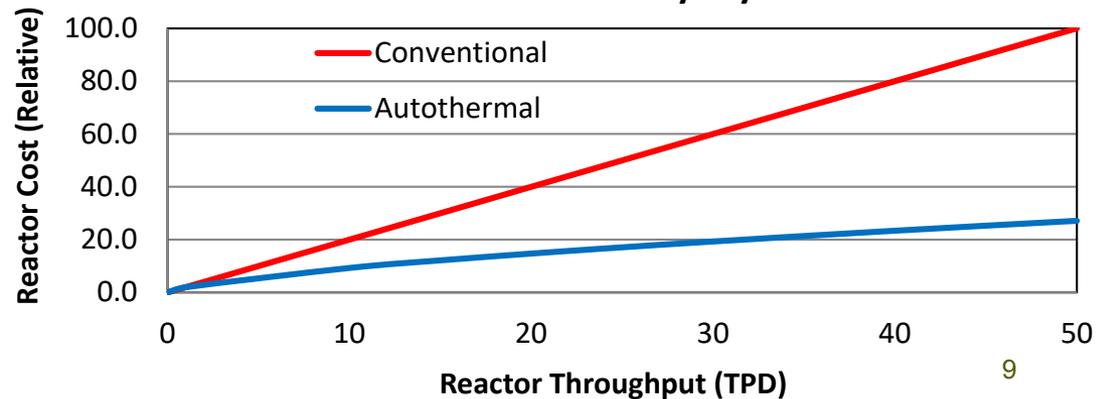
furry



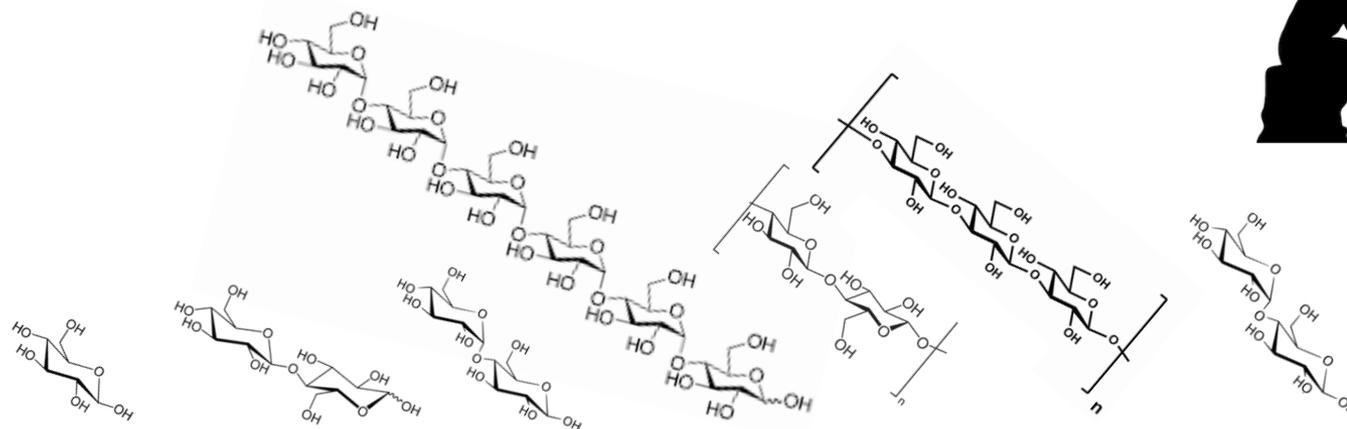
Throughput vs Pyrolyzer Size



Relative Cost of Pyrolyzer



How much sugar can be produced via fast pyrolysis?



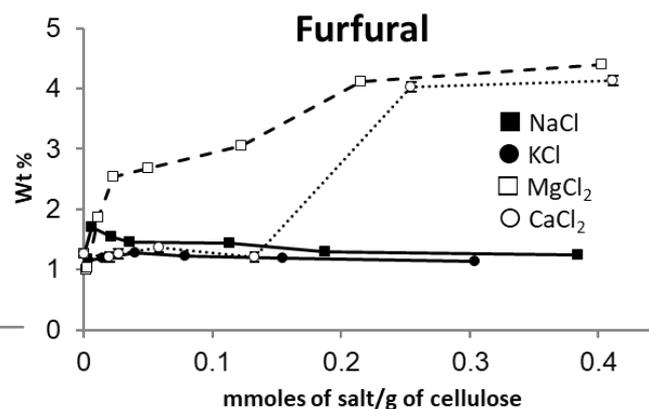
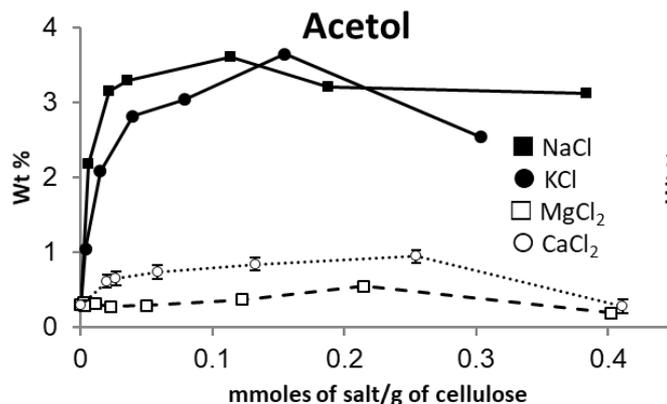
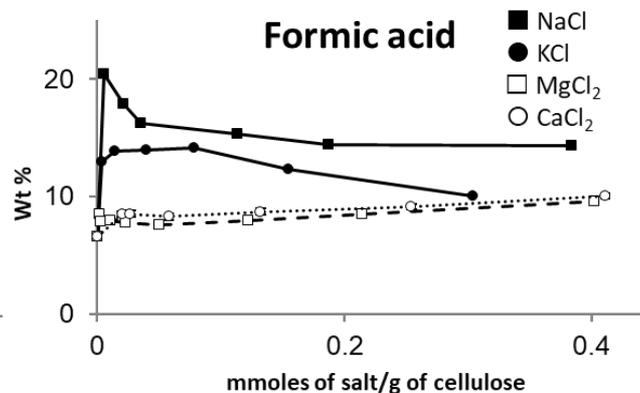
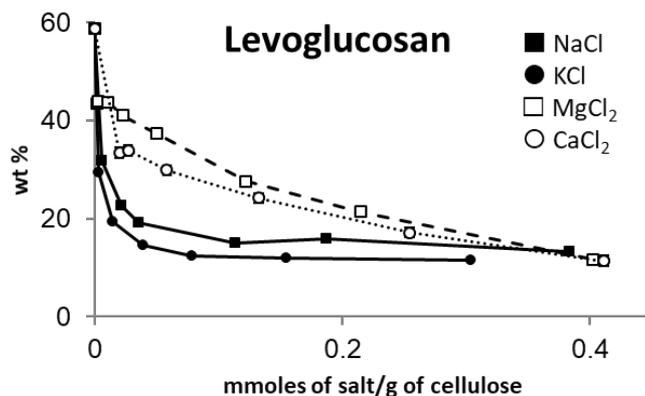
Compound	Glucose (wt%)	Cellobiose (wt%)	Maltose (wt%)	Maltohexaose (wt%)	Cellulose (wt%)	Curdlan (wt%)	Waxy maize starch (wt%)
Levoglucosan	7.00	24.4	20.5	33.1	58.8	44.2	48.5
Unidentified*	40.8	23.6	32.4	20.5	7.1	18.1	13.2

*Includes gases, water and some unidentified organic compounds

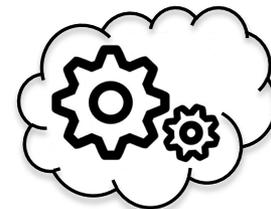
Why don't we find much sugar in bio-oil?



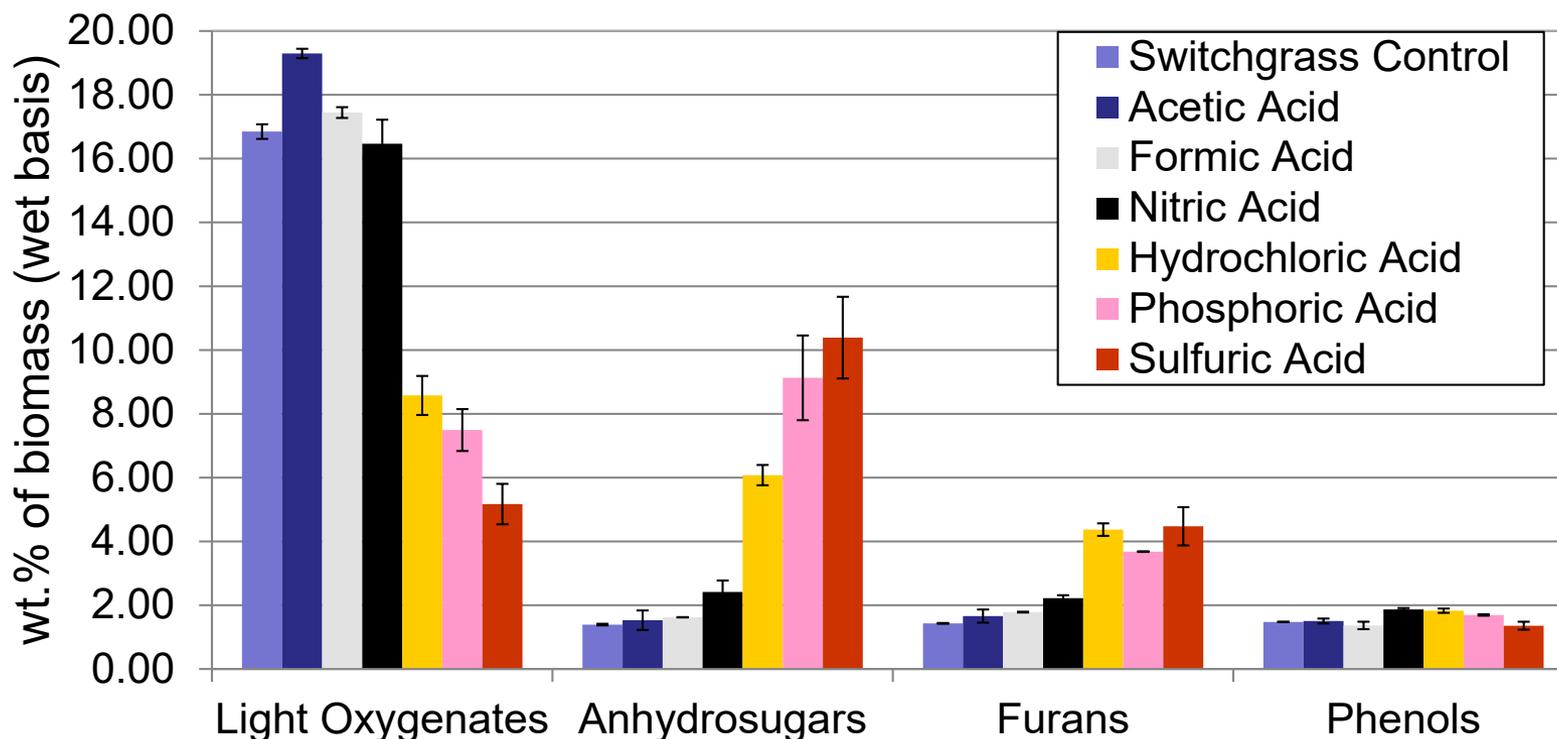
Alkali and alkaline earth metals (AAEM) catalyze pyranose (and furanose) ring fragmentation during pyrolysis



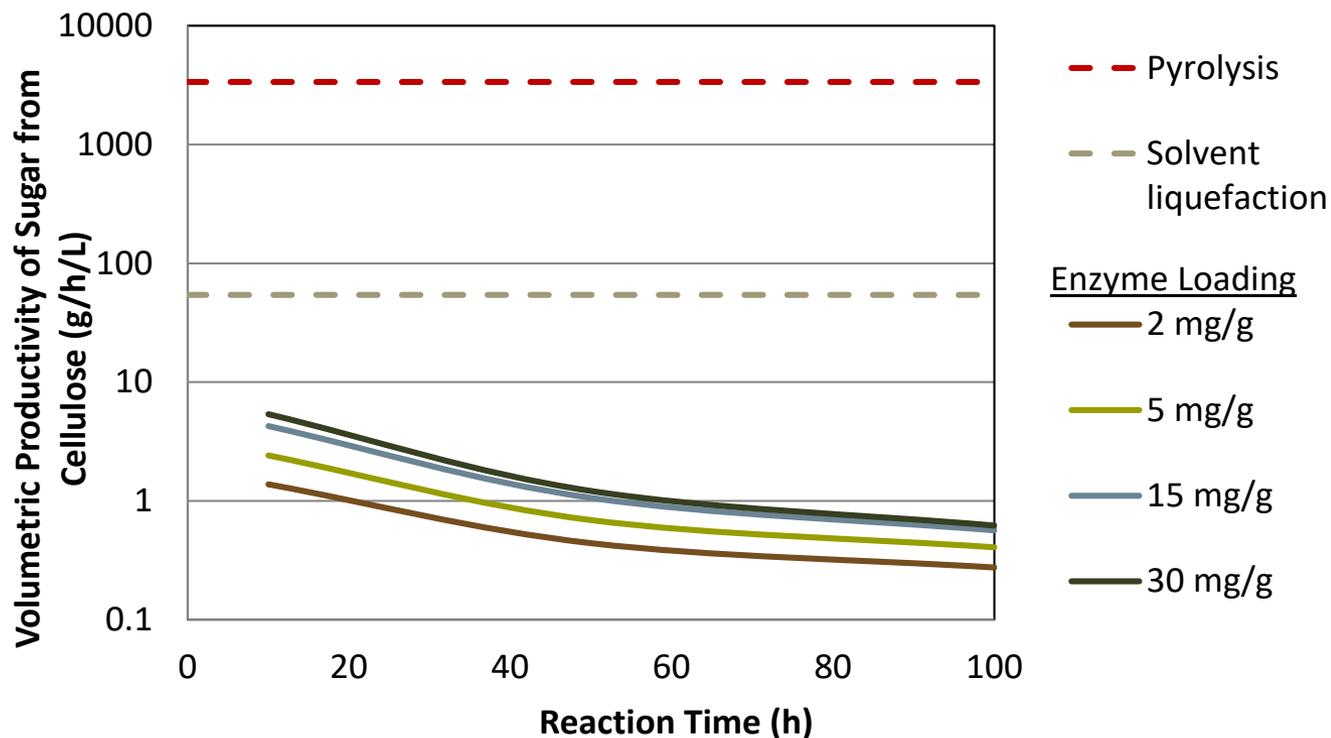
Overcoming the deleterious effect of AAEM on sugar yields



Mineral acid pretreatments convert AAEM into thermally stable salts that passivate catalytic activity of the metals



Process intensification through py sugar production

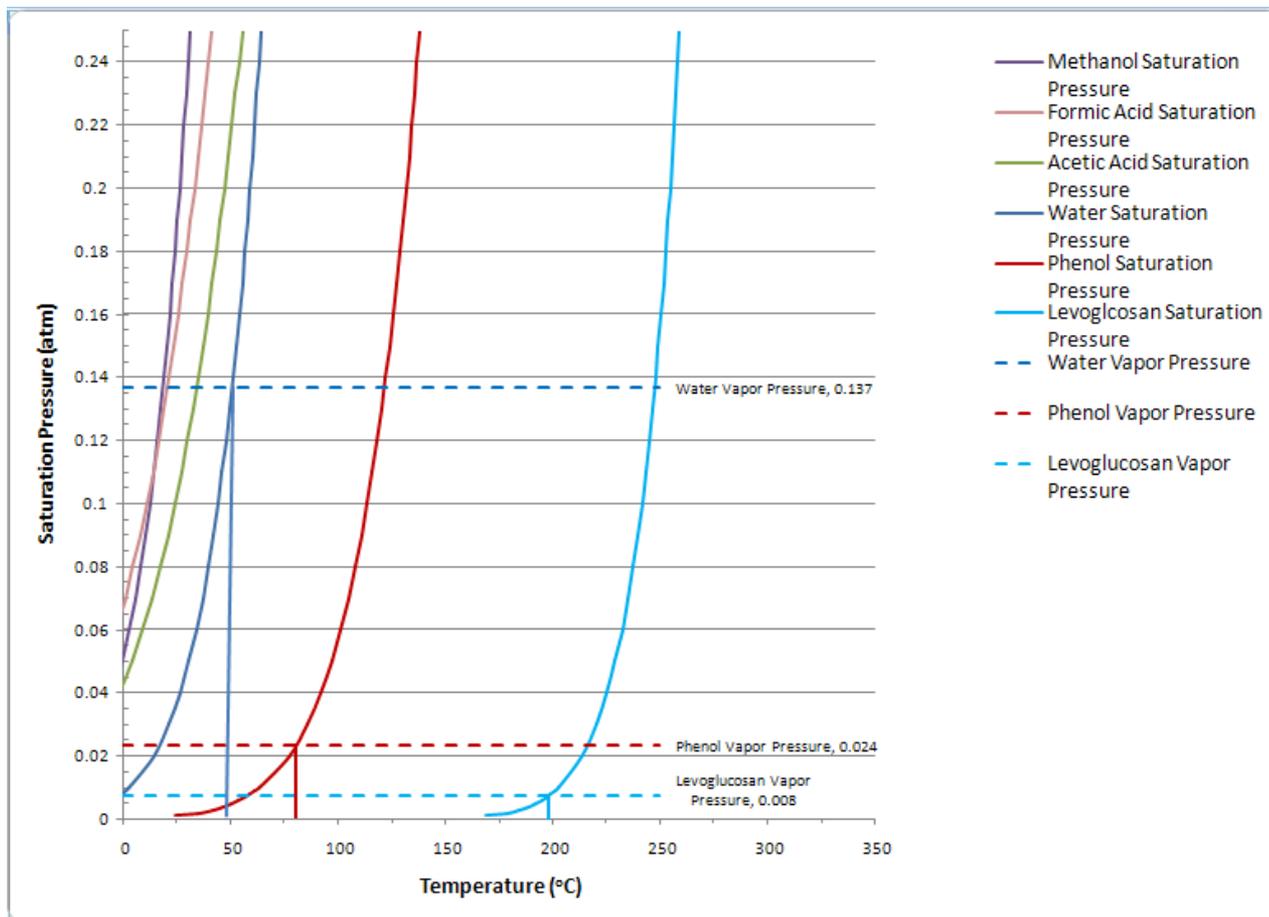


Sugar production for pyrolysis and solvent liquefaction based on sugar yields and processing rates experimentally observed at ISU. Enzymatic hydrolysis data from Nguyen et al., 2015.

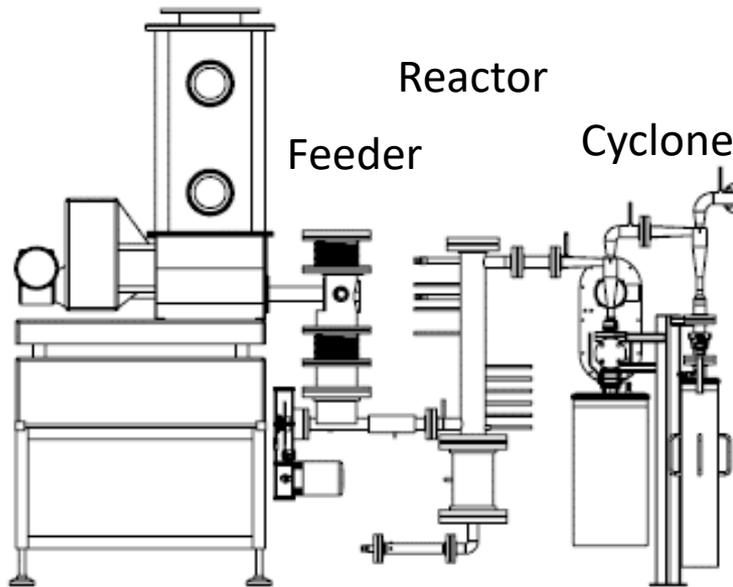
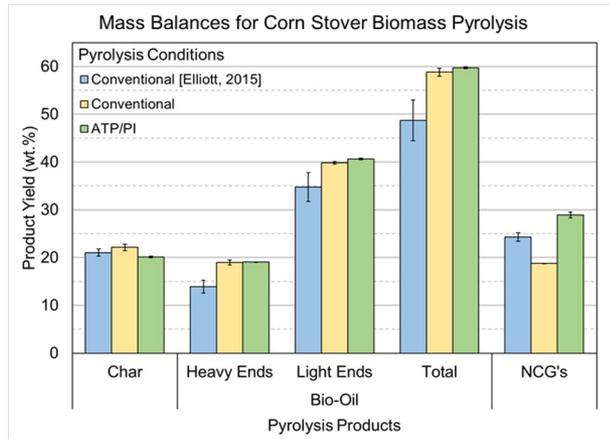
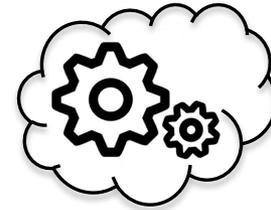
How can we separate organic components of pyrolysis vapors?



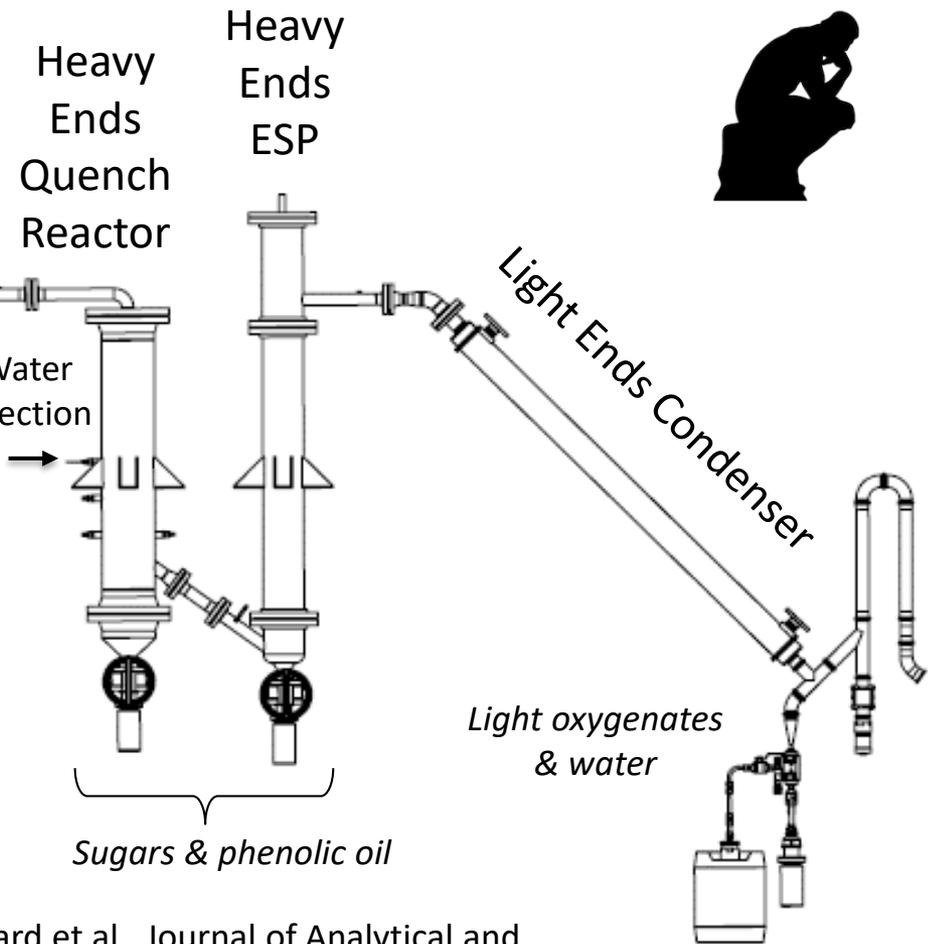
Selective condensation of pyrolysis vapors



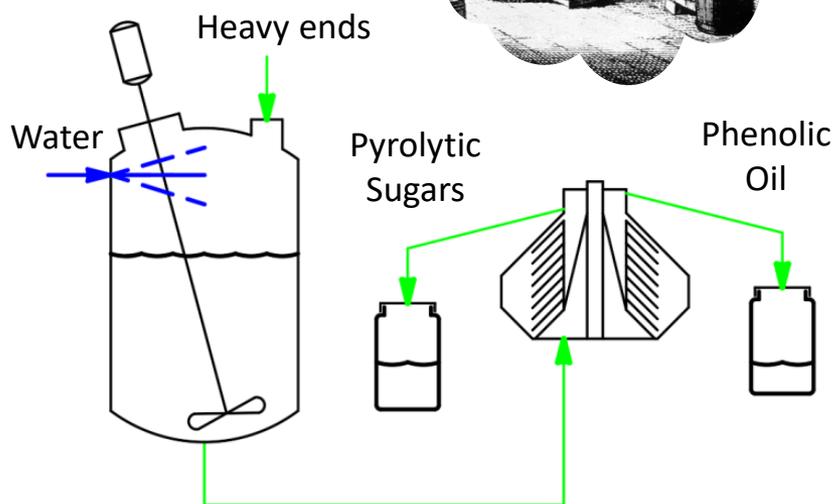
Fractionating Bio-oil Recovery



Biochar



Separating heavy ends into sugars and phenolic oil



Extraction of corn stover-derived heavy ends

Component	Heavy ends (wt% d.b.)	Extracted sugar fraction (wt% d.b.)
Sugars	29.2	61.1
Phenols	61.9	22.4
Acids	3.0	3.8
Other	5.87	12.7

Cleaning py sugars



Pyrolytic Syrup



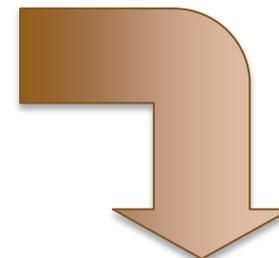
**Contaminant Removal
(phenolic compounds)**



**Pyrolytic Sugars
(99.5% pure)**



**Phenolic
Monomers**

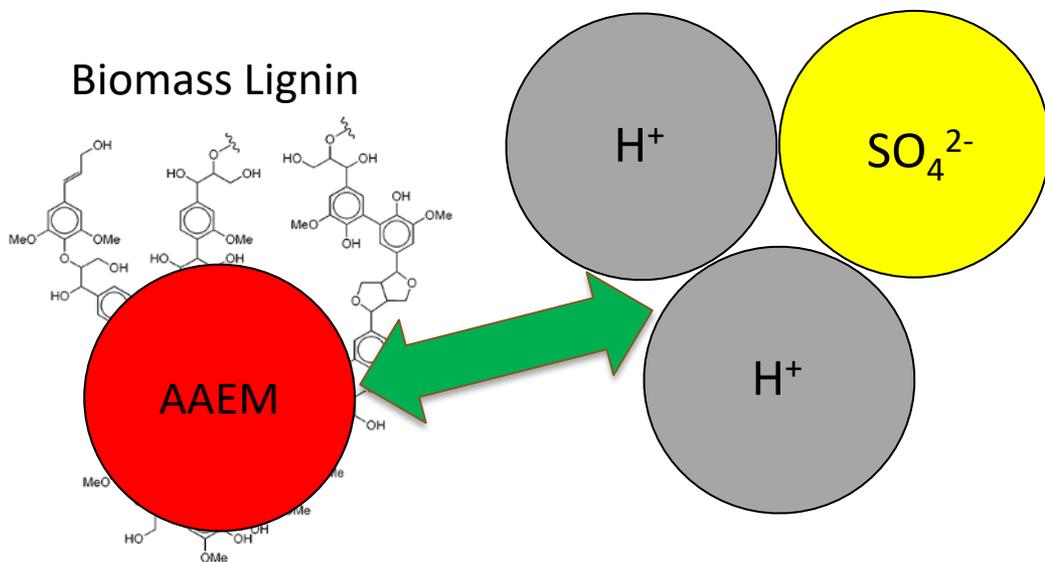


**Levoglucosan
(99.4% pure)**

Why does acid pretreated biomass agglomerate?



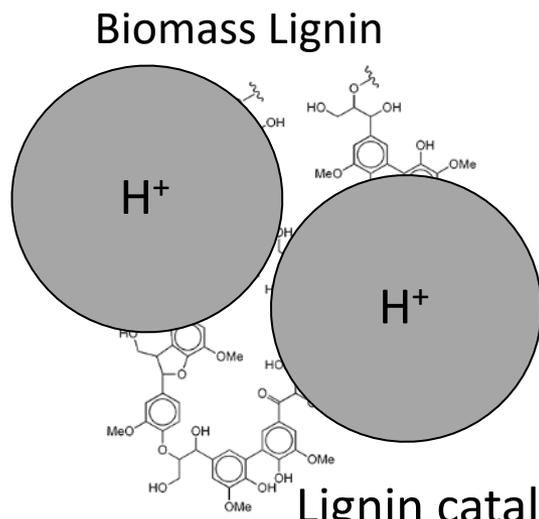
- Alkali and alkaline earth metals in biomass catalyze both pyranose/furanose ring fragmentation and lignin depolymerization
- Lignin melts and agglomerates rather than depolymerizes and volatilizes in the absence of suitable metal catalyst



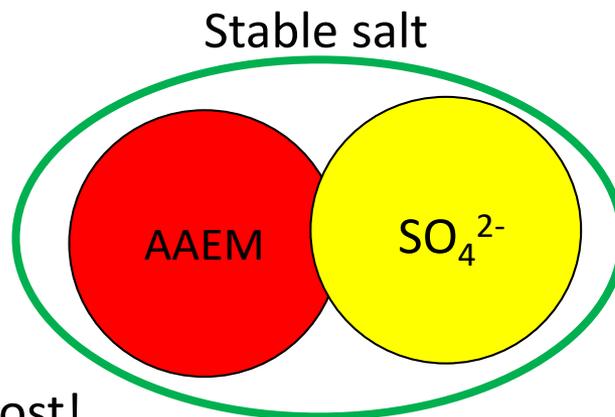
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Lignin catalyst lost!

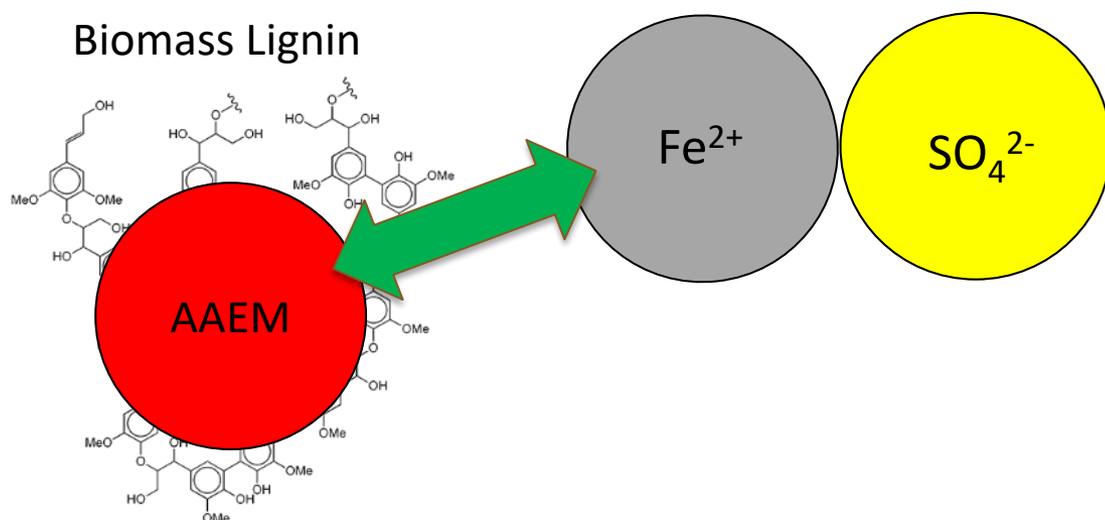


Agglomerated char from acid pretreated biomass

Preventing char agglomeration



- Find an ionic compound for which:
 - Anion passivates AAEM
 - Cation selectively catalyzes lignin depolymerization
- Compound must be water soluble to assure its diffusion into the plant cell walls
- Ferrous sulfate provides these functions

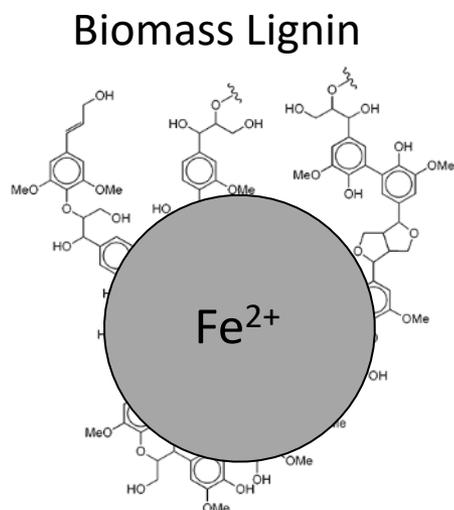


Lignin catalyst replaced!

Preventing char agglomeration

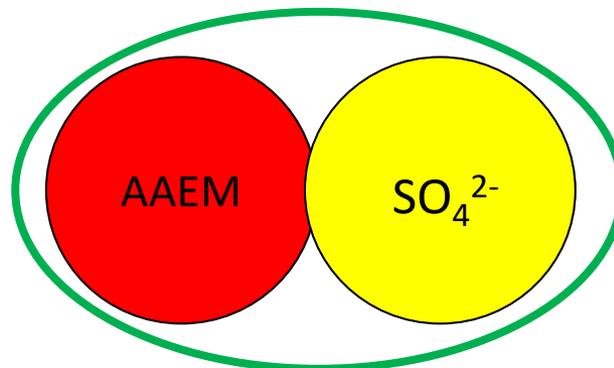


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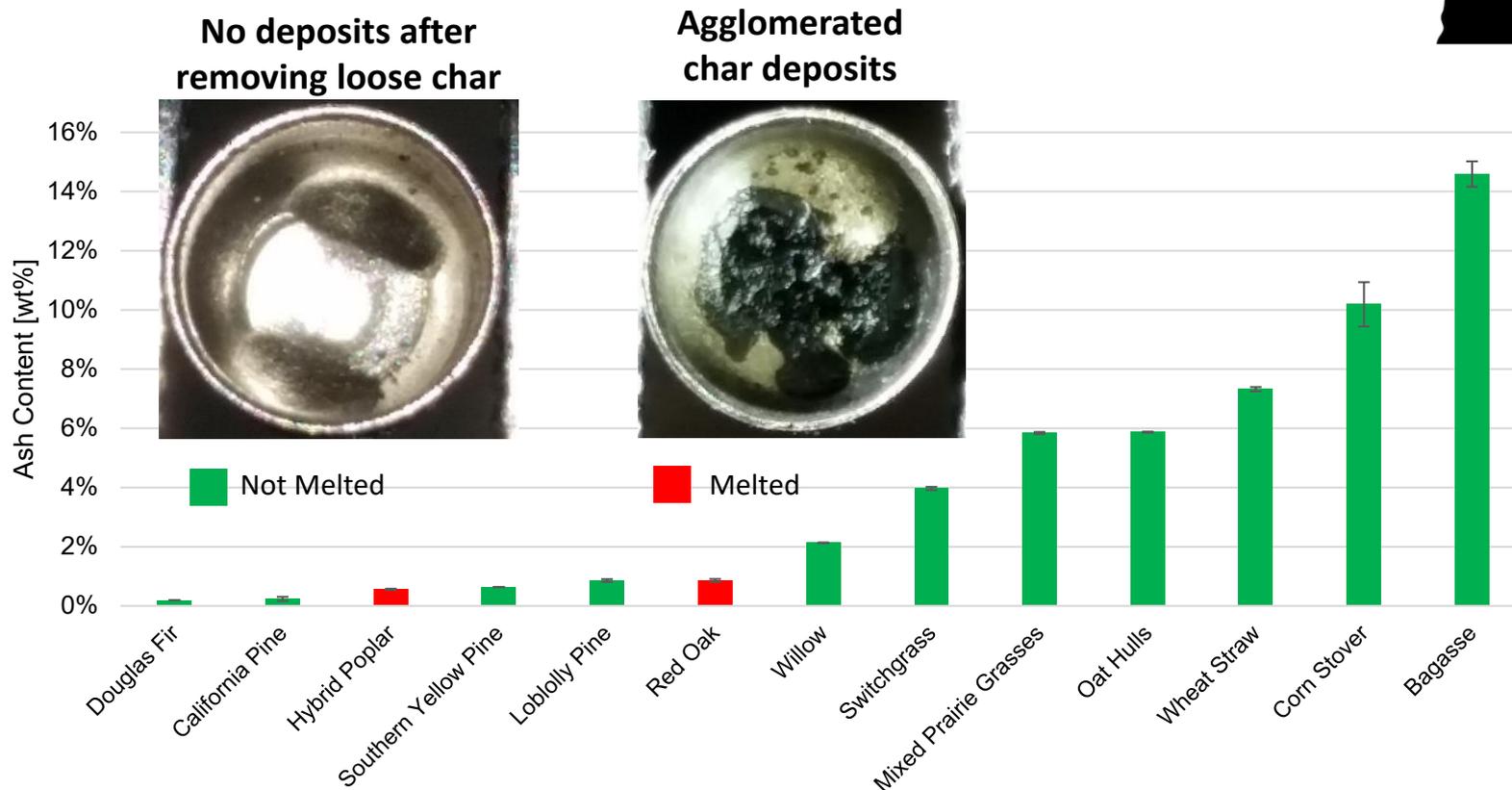
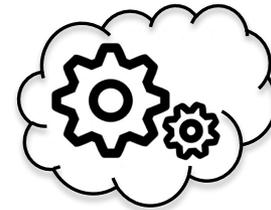
Stable salt



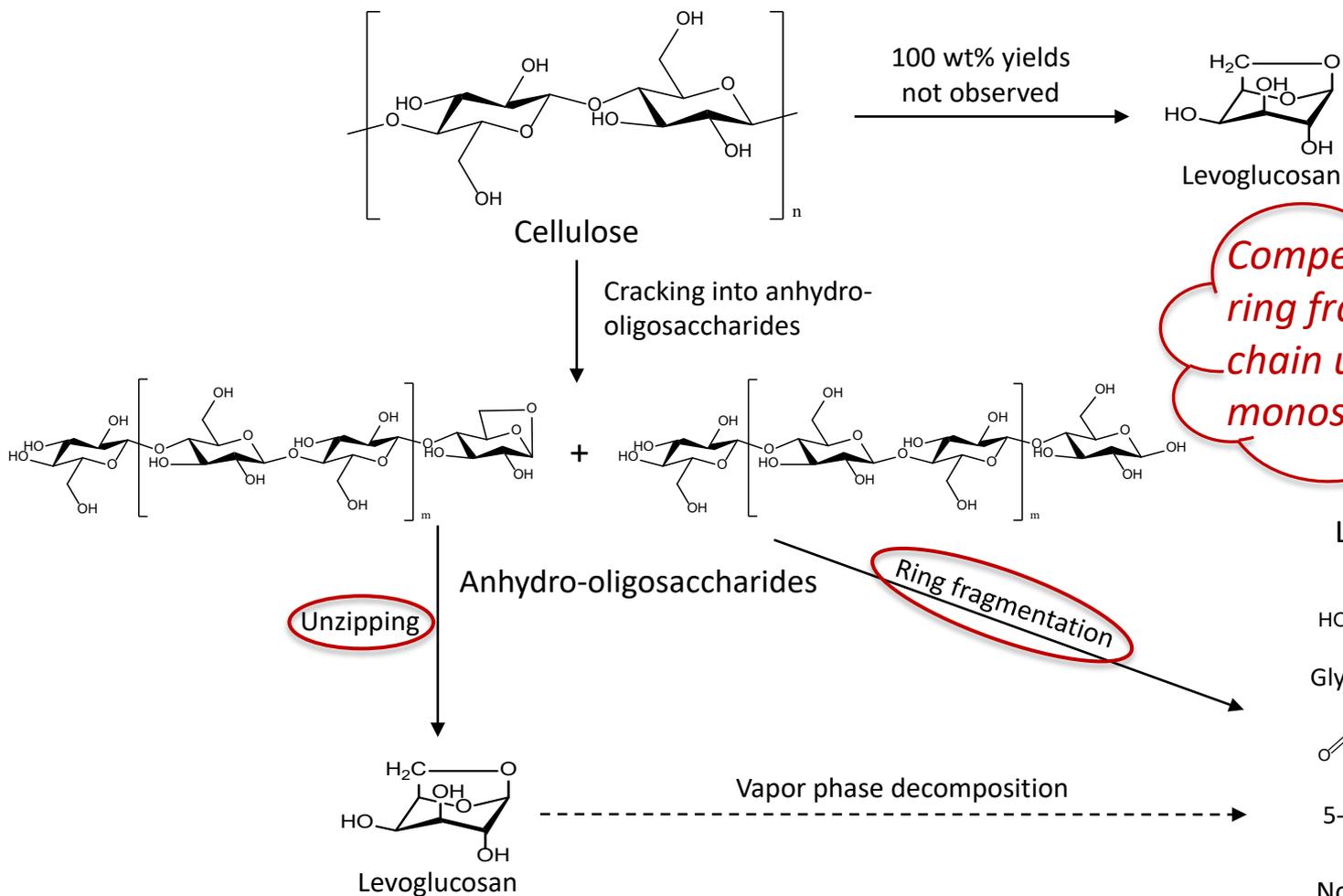
Powdered char from ferrous sulfate pretreated biomass

Preventing char agglomeration

Ferrous sulfate pretreatment eliminated char agglomeration for most kinds of biomass tested



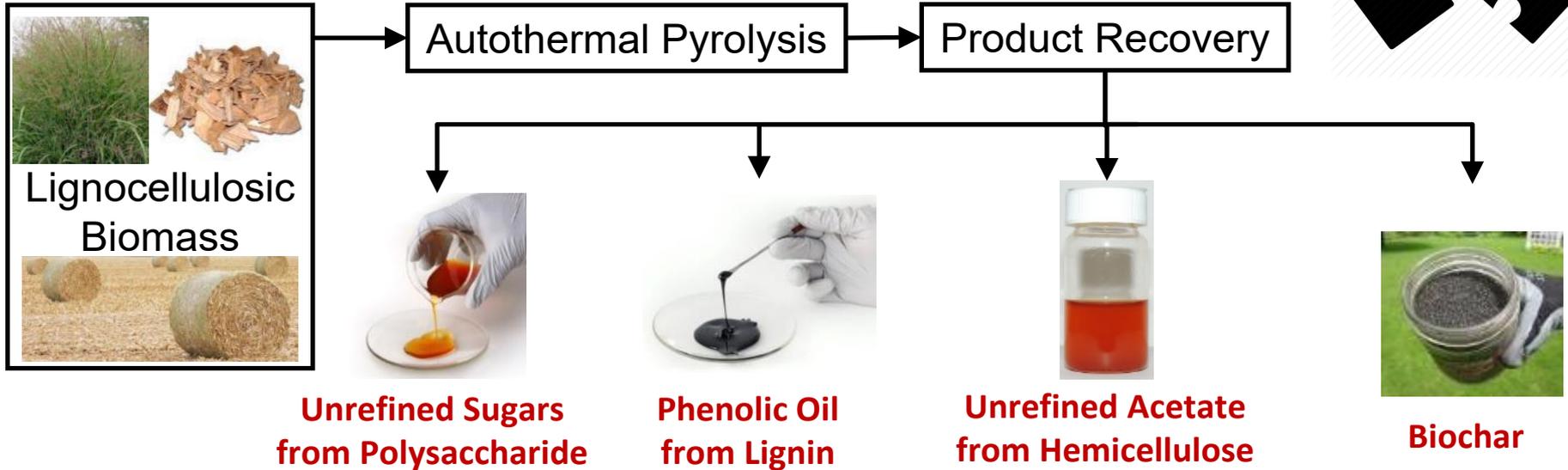
Why don't we get more than 60% yield of sugars from cellulose?



Competition between ring fragmentation and chain unzipping limits monosaccharide yields

- Light Oxygenates
- Glycoaldehyde OCC=O
 - Acetol CC(=O)CO
 - Char O=C1C=CC(=O)O1
 - 5-Hydroxymethyl furfural O=C1C=CC(=O)C=C1CO
 - H₂O
 - Non-condensable gases (eg: CO, CO₂, CH₄)

Putting it all together: Py refinery

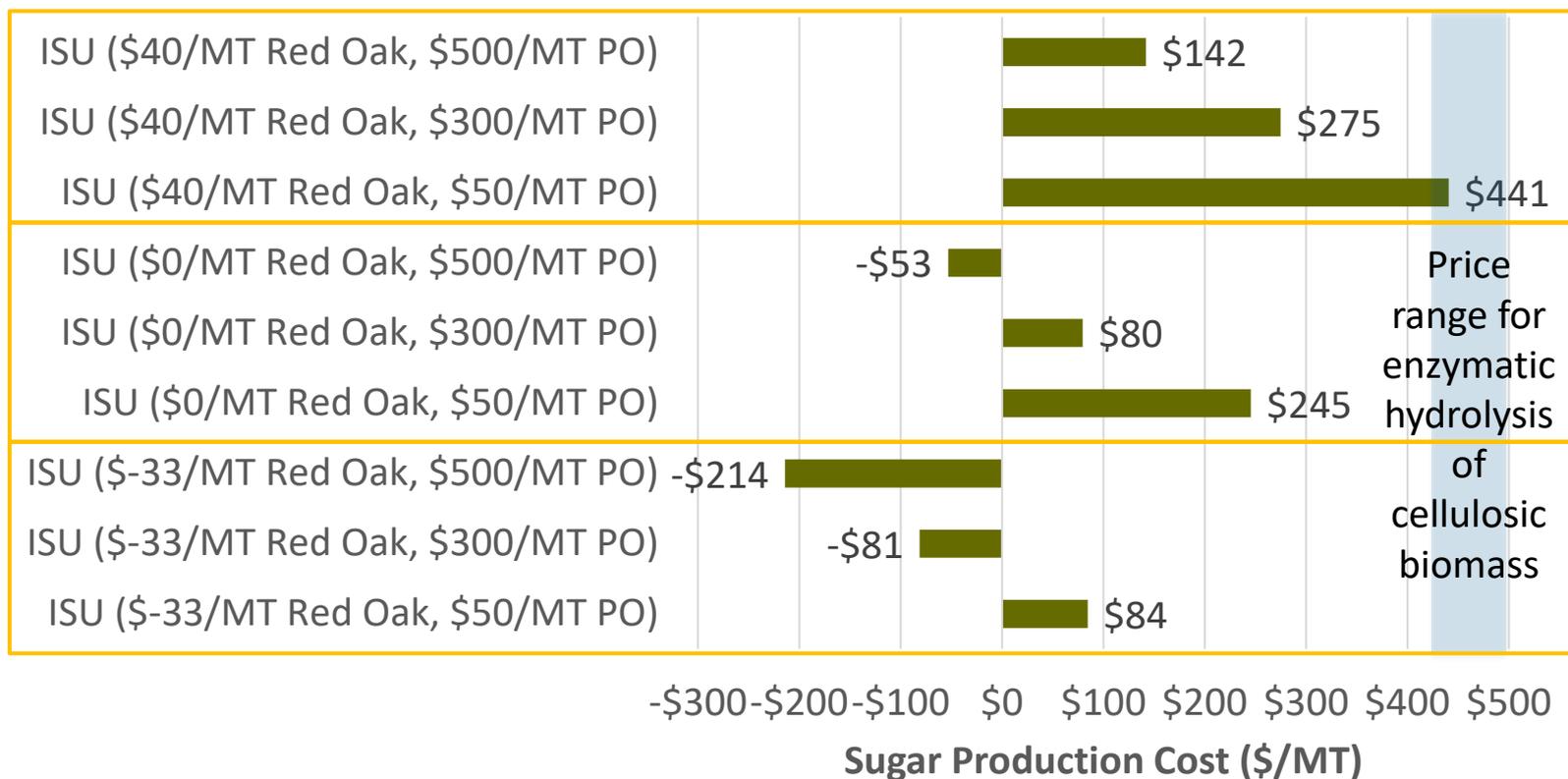


First Generation Products	Ethanol 	Bio-asphalt and marine fuel  	In-plant thermal energy 	Soil Amendment 
Potential Future Products	-Pharmaceuticals -Polymers 	-Octane enhancers -Drop-in Fuels -Biobased Chemicals	-Acetone -Acetic Acid -Bio-cement -Alcohols 	-Activated Carbon 

Economics of Py Refinery



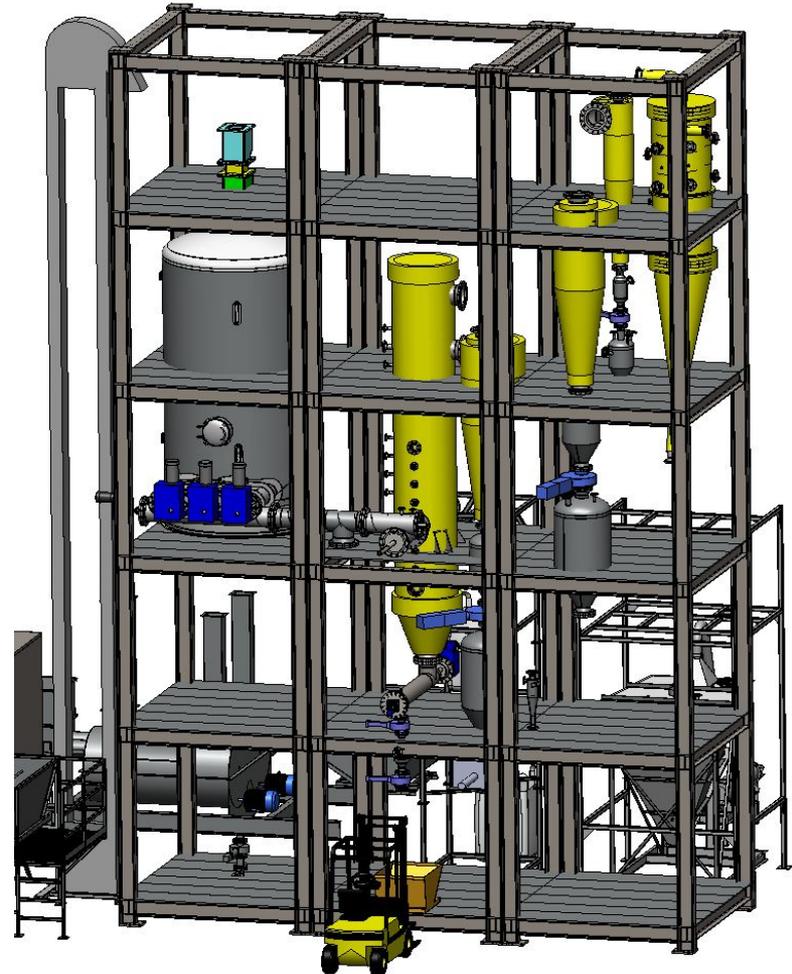
- Low cost cellulosic sugars from woody biomass across a wide range of assumptions
- Feedstock cost and phenolic oil (PO) value are key cost drivers



Next Step: Demonstrate autothermal pyrolysis at 50 tons per day scale



- Iowa project (Redfield, IA)
 - Privately financed by Stine Seed Farms
 - Engineering, Procurement and Construction provided by Frontline Bioenergy
 - Conversion of corn stover into sugars, phenolic oil and biochar
- California project (El Dorado Hills)
 - Funded by CA Energy Commission
 - Partnership with Lawrence Livermore National Laboratory and Frontline Bioenergy
 - Feasibility of converting wood waste into drop-in biofuels



Acknowledgments

I appreciate the creativity and dedicated efforts of a long line of graduate students and staff scientists and engineers who made possible the discoveries and innovations described in my talk.

Heterodoxy in pyrolysis has been made possible through funding from several sources in the last decade including the U.S. DOE, the USDA, the NSF, Conoco-Phillips, Phillips 66, and ExxonMobil. Our work is currently being supported by the U.S. DOE AMO sponsored RAPID Institute as part of an effort to develop modular chemical process intensification (MCPI) in pyrolysis. We are indebted to Stine Seed Company for financing the construction of the first demonstration-scale autothermal pyrolysis plant.

Questions?