



Hydrothermal Liquefaction of Biorefinery Lignin-rich Streams: Carbon Recovery from Residual Aqueous Phase



Stefano Dell'Orco^{1,2}, Edoardo Miliotti², Nolan Wilson³, Andrea Maria Rizzo²,
Kimberly A. Magrini³ and David Chiaramonti^{1,2}

¹ DIEF, University of Florence, Department of Industrial Engineering, Viale Morgagni 40/44, Florence 50139, Italy

² Renewable Energy Consortium for Research and Demonstration – RE-CORD, Viale Kennedy 184, 50038 Scarperia, Florence, Italy.

³ National Renewable Energy Laboratory – NREL, 15013 Denver West Parkway, Golden, CO 80401-3393, USA

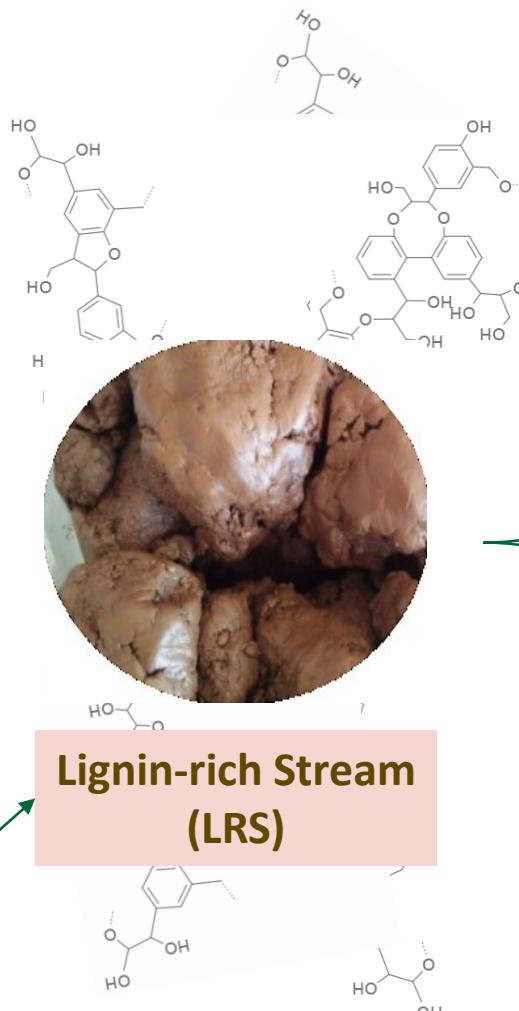
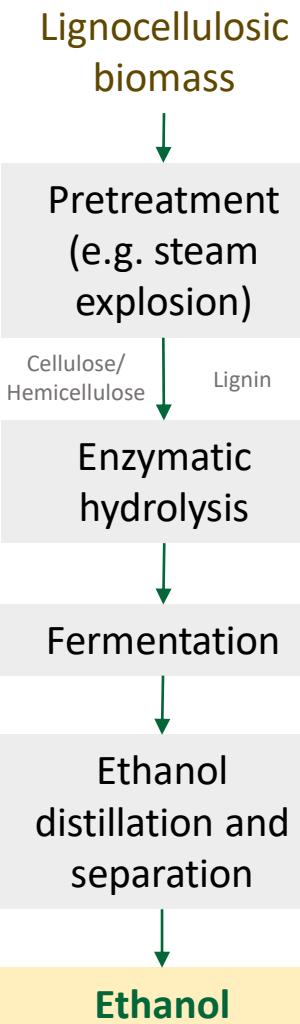


Summary

- Introduction
- Hydrothermal liquefaction of Lignin-rich streams
- Residual aqueous phase collection and characterization
- Aqueous phase valorization: Liquid-Liquid Extraction
- Conclusion and further developments



Ethanol biorefinery: Lignin-rich streams



Parameter	Value	Unit
Total lignin	57.8	wt.% daf
Structural sugars	35.5	wt.% daf
Moisture	69.7	wt.% wb
Volatile matter	71.0	wt.% db
Ash	2.6	wt.% db
Fixed carbon	26.4	wt.% db
C	54.2	wt.% db
H	5.9	wt.% db
N	1.0	wt.% db
S	0.2	wt.% db
O*	36.1	wt.% db
Slurry pH	5	-



Hydrothermal Liquefaction of LRS

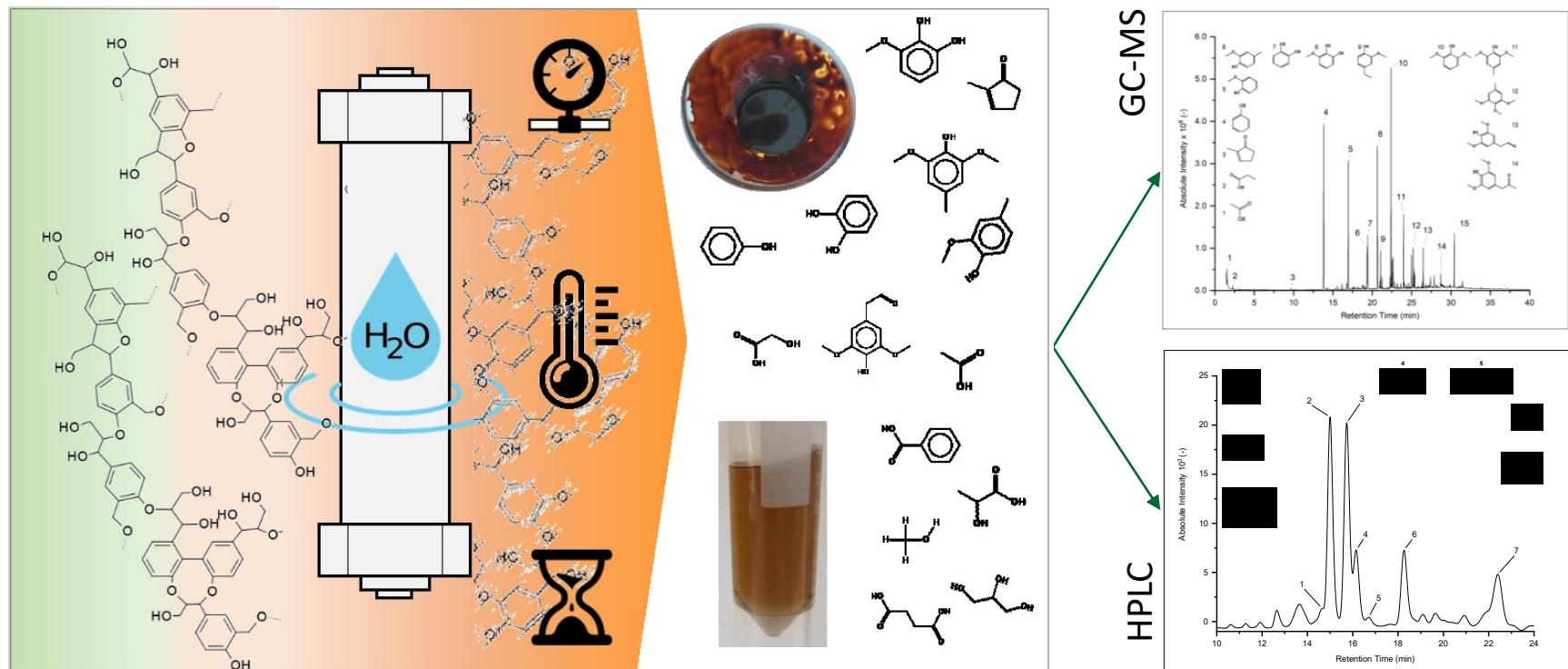


Micro-reactors test bench:

- Tubular batch reactor, OD 3/4"
- Fluidized sand bath
- Fast water bath cooling

HTL reaction conditions:

- Temperature = 350° C
- Residence time = 10 min
- Lignin to water ratio = 10 wt.%

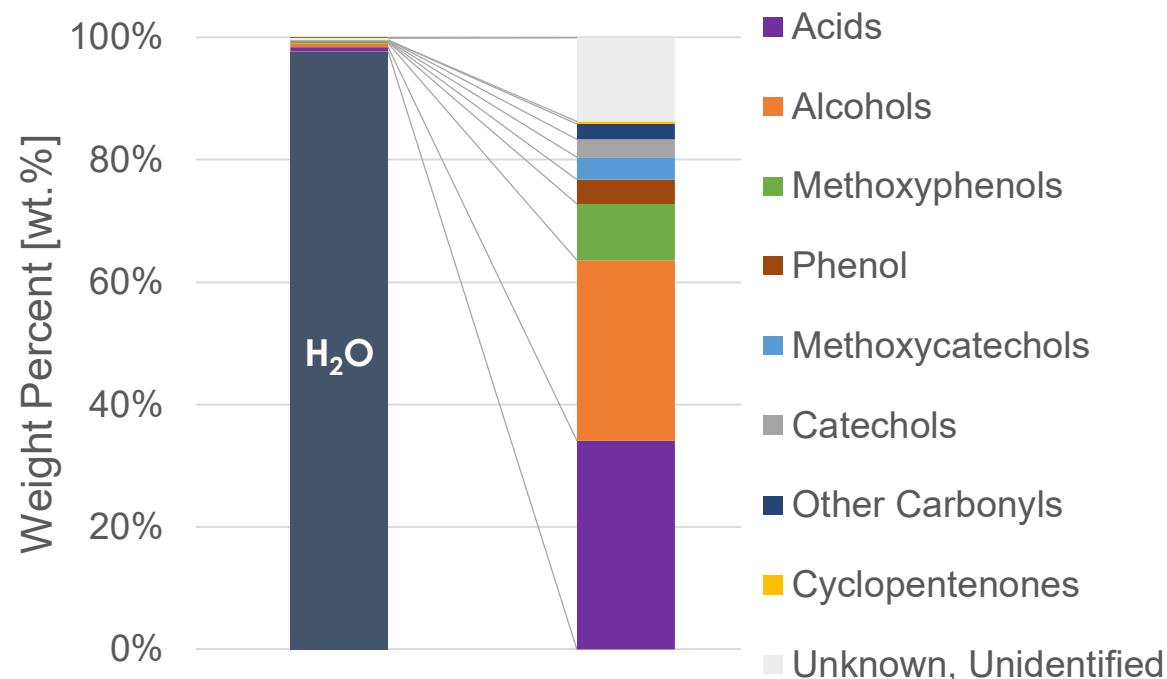




LRS-HTL aqueous phase characterization

Analysis	Value	Units
TOC	9.73 ± 0.780	g/l
Water content KF Titration	97.7 ± 1.09	wt.%
ICP – Inorganic elements*	0.114	wt.%
pH	4.27	-

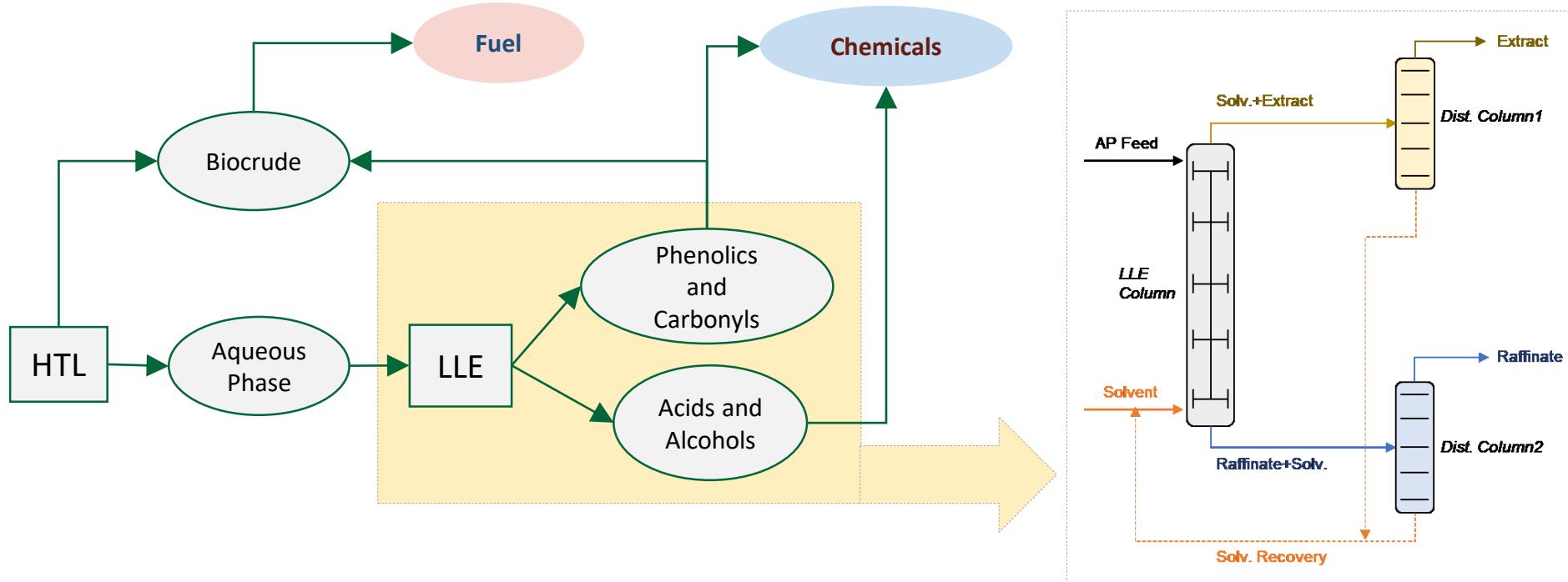
*Na, K, Ca, Fe, Al, Mg, Zn



- Total detected organics → **1.93 wt.%** → *GC-MS, HPLC*
- Organic mass balance closure → **83-86 %** → *KF, TOC*
 - Acids (34 wt.%) and Alcohols (30 wt.%) → HPLC
 - Phenolics (20 wt.%) and Carbonyls (3.0 wt.%) → GC-MS



Liquid-Liquid Extraction HTL integration



- Liquid-Liquid Extraction (LLE) → Selective separation of phenolics compounds
- Potential valorization pathways:
 - Biocrude yield enhancement for Fuel production
 - Precursors for Chemicals and Bioproducts



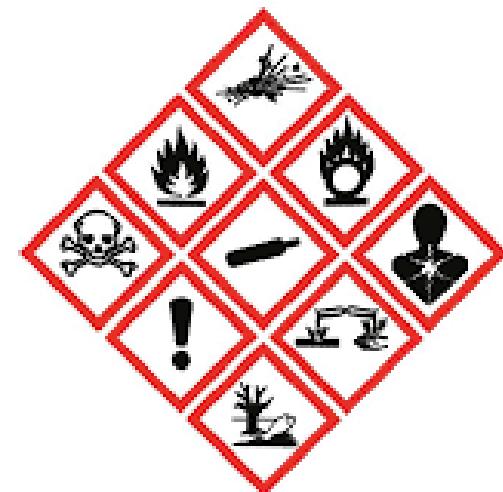
Solvent selection

ID	Name	DH_vap [kJ mol-1]	T_boil [°C]	Flash Point [°C]	Solubility in H2O [g/l]	Price* [\$ kg-1]
DEE	Diethyl ether	28	34.5	-45	60.5	2.0
EtAc	Ethyl Acetate	33	77	-4	83	0.8
BuAc	Butyl Acetate	41	126	22	6.8	1.7
MIBK	Methyl Isobutyl ketone	40.5	117	13	19.1	1.4

*Updated using producer price index to adjust values

ID	Safety score	Healthy score	Env. Score	Ranking
DEE	10	3	7	Highly Hazardous
EtAc	5	3	3	Recommended
BuAc	4	2	3	Recommended
MIBK	4	2	3	Recommended

D. Prat, A. Wells, J. Hayler, H. Sneddon, C.R. McElroy, S. Abou-Shehada, P.J. Dunn, CHEM21 selection guide of classical- and less classical-solvents, Green Chem. 18 (2015)





Liquid-Liquid Extraction tests

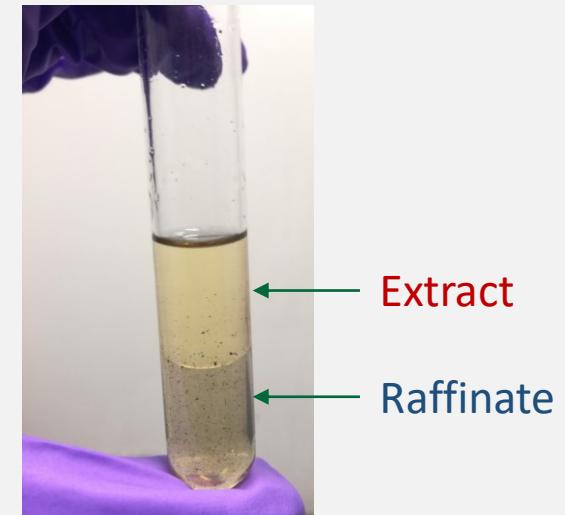
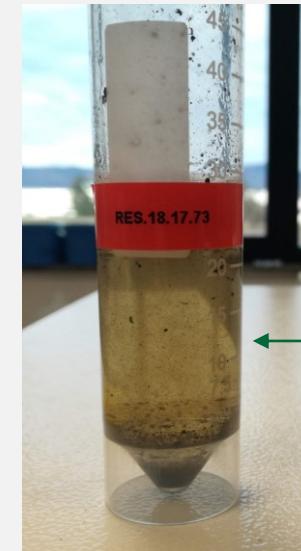
- Lab scale solvents screening tests

- Experimental conditions

- Constant AP feed pH
- Temperature 25°C
- Extraction time 5 min
- Phase separation after 5 min of centrifugation
- Solvent to AP ratio = 1

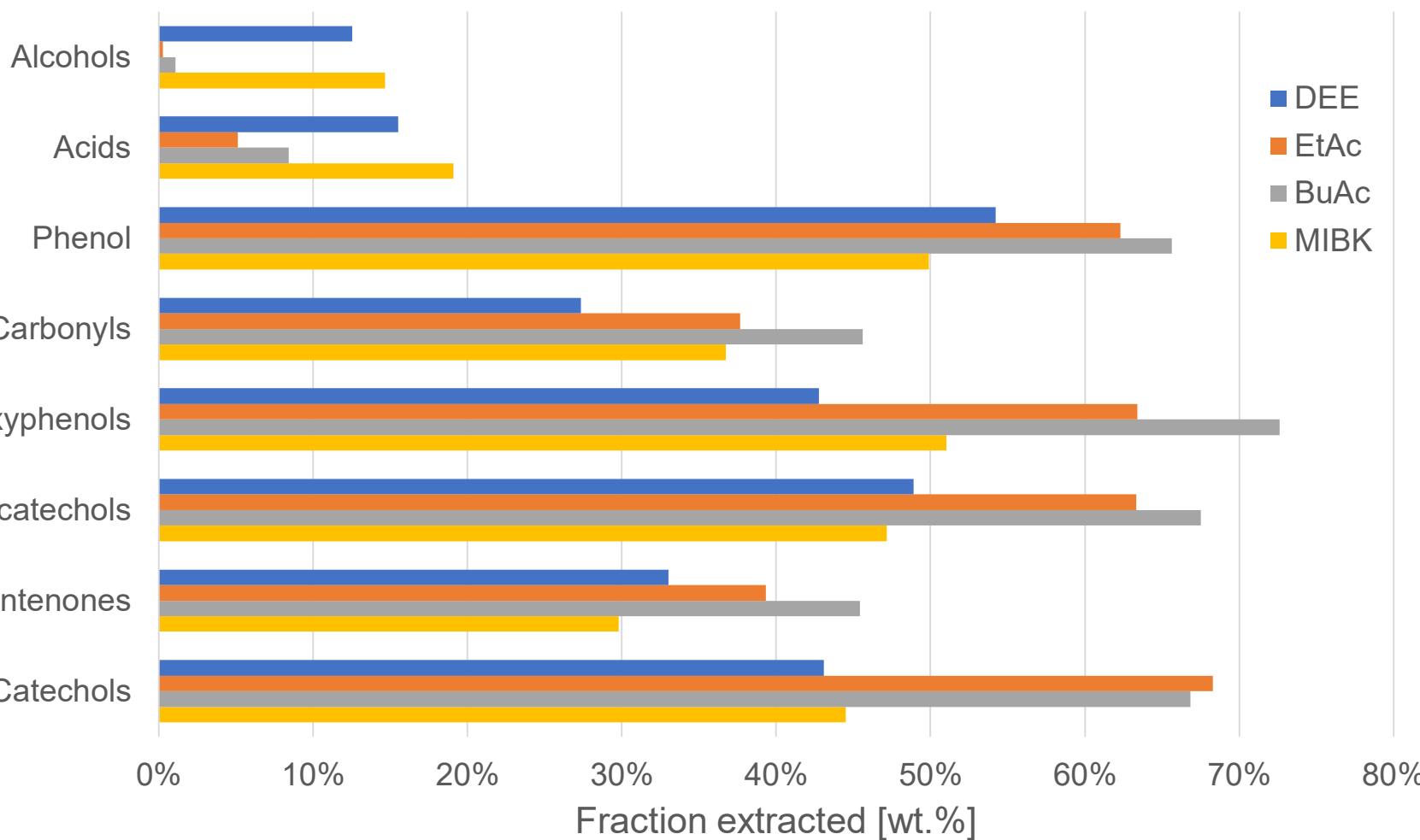
- Analytical methods

- AP feed: GC-MS and HPLC
- Extract: GC-MS
- Raffinate: HPLC





Liquid-Liquid Extraction: Extract recovery

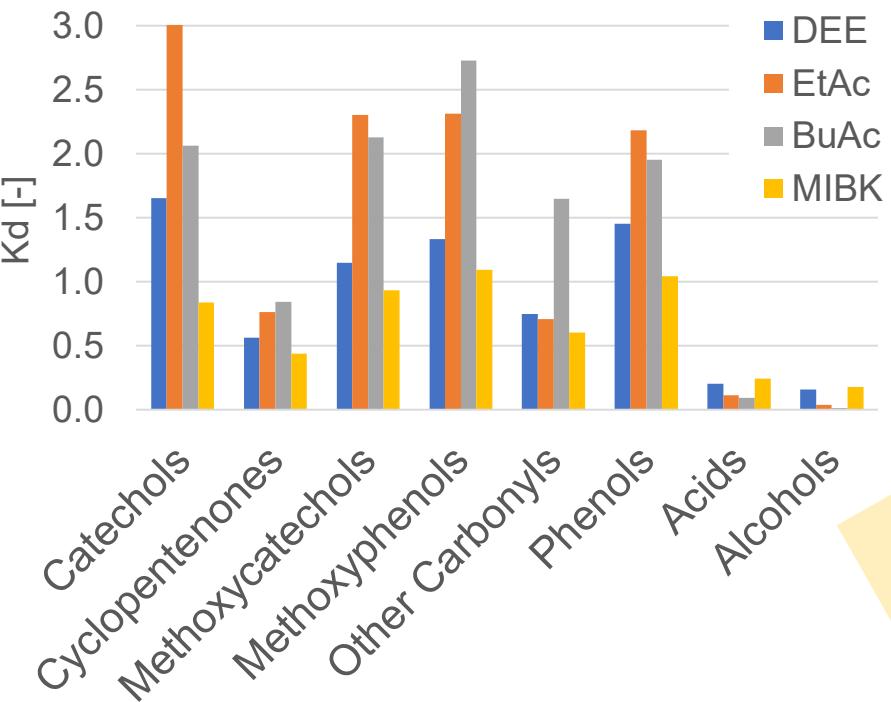


- Fraction Extracted $\rightarrow F_i = \frac{m_{i,org}}{m_{i,feed}}$

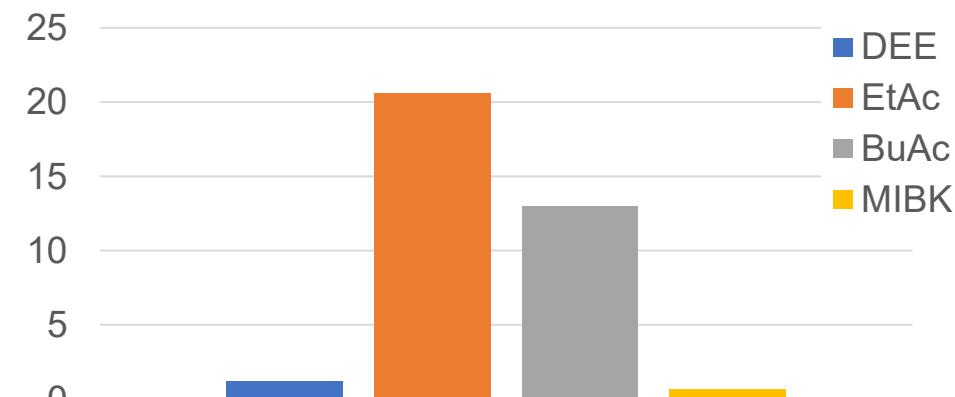


Liquid-Liquid Extraction: K_d and Selectivity

Distribution ratio



Selectivity Phenolics/Acids&Alcohols



- Distribution ratio $\rightarrow K_{d,i} = c_{extr,i}/c_{raff,i}$
- Selectivity $\rightarrow \alpha_{P,A} = K_{d,P}/K_{d,A}$

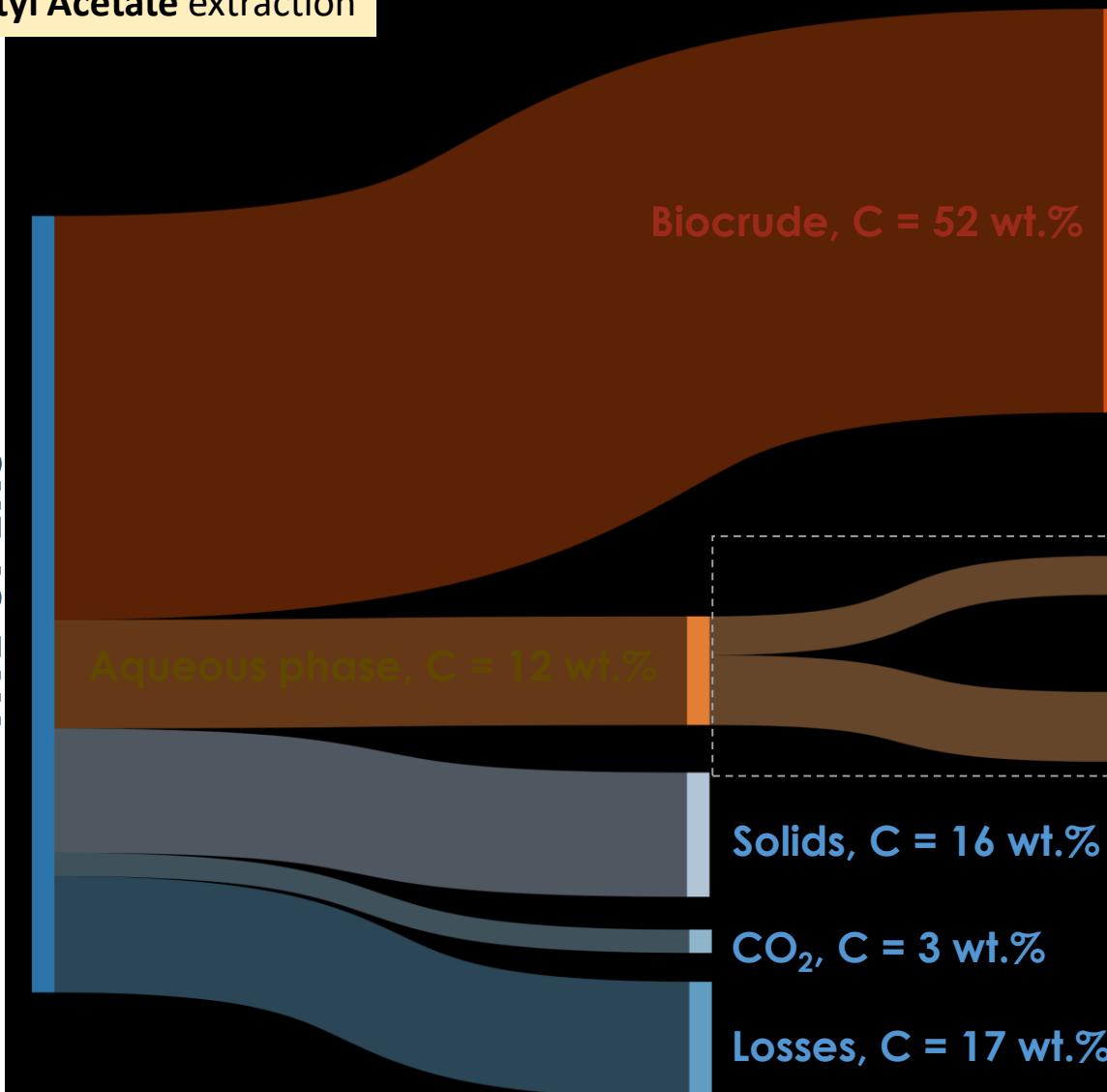
Highest phenolics K_d and selectivity compared to acids and alcohols with **Ethyl acetate** and **Butyl acetate**



Li

Butyl Acetate extraction

HTL of LRS



- 83% phenolics
- 16% acids

LLE

- 13% phenolics
- 46% acids
- 41% alcohols



LLE integration for biocrude enhancement

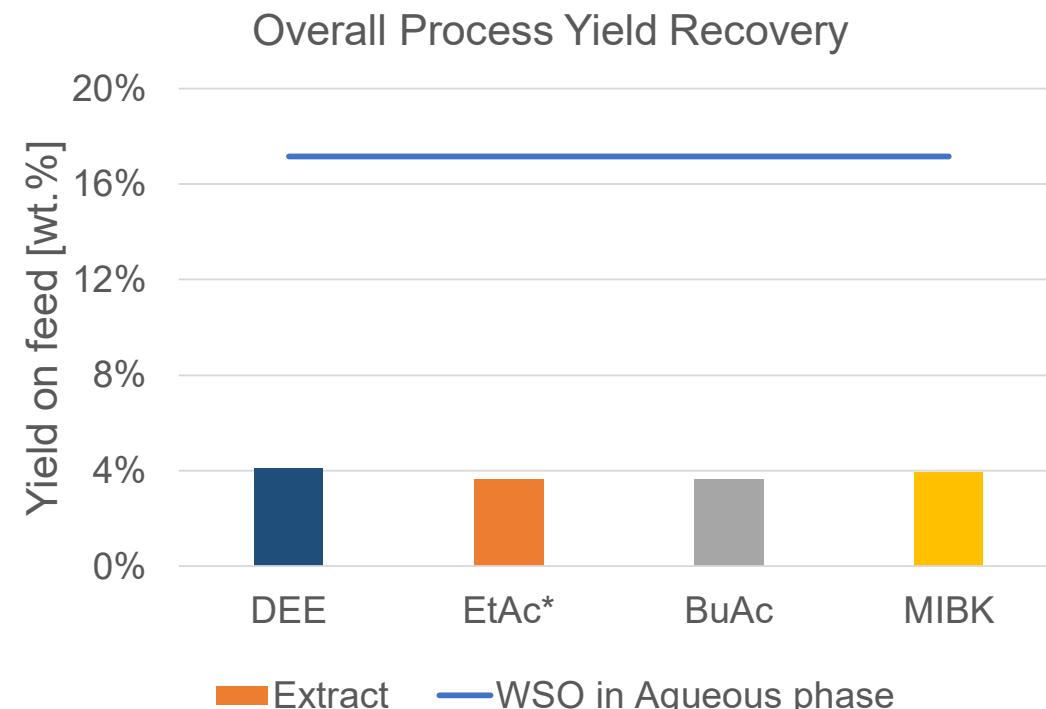


HTL → AP → LLE

Extract

Biocrude

Fuel



- About 8 % of increment in **Biocrude yield**
- No differences in extract yields changing the solvent
- DEE and MIBK extracts higher fractions of acids

*estimated acetic, propionic acids and methanol content from BuAc data



LLE integration for chemicals precursors



HTL → AP → LLE

Extract

Separation
Purification

Chemical
precursors



Chemicals

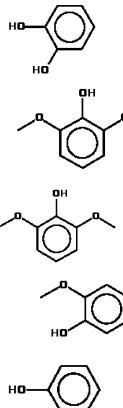


- Lignocellulosic Ethanol Biorefinery

- Estimated lignin amount produced in EU: from **49 kTA (2017)** to **4340 kTA (2030)***

* Data estimated from ethanol data: C. Chudziak, G. Alberts, A. Bauen, Ramp up of lignocellulosic ethanol in Europe to 2030 Final Report, 2017, E4tech

Compound <i>extracted with EtAc</i>	Kd	Fraction extracted [wt.%]	Yield on feed * [wt.%]	Production 2017 [kTA]	Production 2030 [kTA]	Market size [kTA]	Price [\$ kg ⁻¹]
1,2-Benzenediol	2.5	71%	0.02%	0.01	0.9	44 ^a	4.0 ^b
1,2-Benzenediol, 3-methoxy-	2.3	64%	0.05%	0.02	2.2	-	-
Phenol, 2,6-dimethoxy-	3.6	78%	0.06%	0.03	2.4	40 ^c	2.0 ^c
Phenol, 2-methoxy-	1.5	61%	0.03%	0.01	1.2		
Phenol	2.2	69%	0.05%	0.03	2.4	11400 ^b	1.1 ^b



* Estimation of compound purification efficiency of 10 %

^a <https://www.gep.com/mind/blog/catechol-faces-supply-crunch-while-prices-rise>

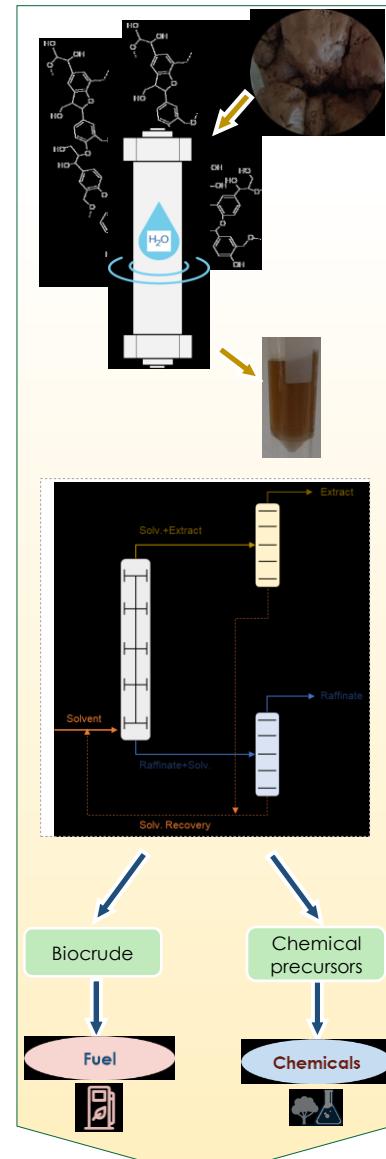
^b A.N. Wilson, A. Dutta, B.A. Black, C. Mukarakate, J.A. Schaidle, W.E. Michener, G.T. Beckham, M.R. Nimlos, Valorization of Aqueous Waste Streams from Thermochemical Biorefineries, Green Chem. (2019) Accepted. doi:10.1039/c9gc00902g.

^c P. Varanasi, P. Singh, M. Auer, P.D. Adams, B.A. Simmons, S. Singh, Survey of renewable chemicals produced from lignocellulosic biomass during ionic liquid pretreatment, Biotechnol. Biofuels. 6 (2013) 1. doi:10.1186/1754-6834-6-14.



Conclusions

- AP from LRS-HTL still contains **2-3 wt.% of organic matters**
 - 12 wt.% of C trapped in AP
- Liquid-liquid extraction (LLE) was evaluated as possible pathway for **selective organics recovery**
- DEE, EtAc, BuAc and MIBK were tested as extraction solvent
 - EtAc and BuAc → highest selectivity for phenolics extraction
 - 3 wt.% of C recovered in extract
- LLE-Integration for biocrude yield enhancement
 - Potential biocrude yield increment of 8 %
- LLE-Integration for bio-based chemicals precursors
 - Targeted molecules: 1,2 benzenediol, 1,4-Benzenediol, 2-methoxy-Phenol, 2,6-dimethoxy-, Phenol, 2-methoxy-.





Thanks for the attention!

Stefano Dell'Orco ^{1,2}, Edoardo Miliotti ², Nolan Wilson ³,
Andrea Maria Rizzo ², Kimberly A. Magrini ³ and David
Chiaramonti ^{1,2}

¹ **DIEF, University of Florence**, Department of Industrial Engineering, Viale Morgagni 40/44, Florence 50139, Italy

² **Renewable Energy Consortium for Research and Demonstration – RE-CORD**, Viale Kennedy 184, 50038 Scarperia, Florence, Italy.

³ **National Renewable Energy Laboratory – NREL**, 15013 Denver West Parkway, Golden, CO 80401-3393, USA

